

**CHAPTER 7 REVIEW***Chemical Formulas and Chemical Compounds***SECTION 7-1****SHORT ANSWER** Answer the following questions in the space provided.

- 1.** C In a Stock name such as iron(III) sulfate, the roman numeral tells us \_\_\_\_\_.
  - (a) how many atoms of Fe are in one formula unit
  - (b) how many sulfate ions can be attached to the iron atom
  - (c) the charge on each Fe ion
  - (d) the total positive charge of the formula unit
- 2.** C The result of changing a subscript in a correctly written chemical formula is to \_\_\_\_\_.
  - (a) change the number of moles represented by the formula
  - (b) change the charges on the other ions in the compound
  - (c) change the formula so that it no longer represents the compound it previously represented
  - (d) have no effect on the formula
- 3.** The explosive TNT has the molecular formula  $C_7H_5(NO_2)_3$ .
 

<u>4 elements</u>	<b>a.</b> How many elements make up this compound?
<u>6 oxygen atoms</u>	<b>b.</b> How many oxygen atoms are present in one molecule of $C_7H_5(NO_2)_3$ ?
<u>21 atoms</u>	<b>c.</b> How many atoms in total are present in one molecule of $C_7H_5(NO_2)_3$ ?
<u><math>4.2 \times 10^{24}</math> atoms</u>	<b>d.</b> How many atoms are present in a sample of $2 \times 10^{23}$ molecules of $C_7H_5(NO_2)_3$ ?
- 4.** How many atoms are present in each of these formula units?
 

<u>11 atoms</u>	<b>a.</b> $Ca(HCO_3)_2$
<u>45 atoms</u>	<b>b.</b> $C_{12}H_{22}O_{11}$
<u>10 atoms</u>	<b>c.</b> $Fe(ClO_2)_3$
<u>9 atoms</u>	<b>d.</b> $Fe(ClO_3)_3$
- 5.**

<u><math>N_2O_5</math></u>	<b>a.</b> What is the formula for the compound dinitrogen pentoxide?
<u>carbon(IV) sulfide</u>	<b>b.</b> What is the Stock name for the covalent compound $CS_2$ ?
<u><math>H_2SO_3</math></u>	<b>c.</b> What is the formula for sulfurous acid?
<u>phosphoric acid</u>	<b>d.</b> What is the name for the acid $H_3PO_4$ ?

# Guided And Study Chemical Names And Formulas

**Jin-Ying Zhang**



## **Guided And Study Chemical Names And Formulas:**

**CHEMISTRY HANDBOOK & STUDY GUIDE Gr11-12 NE** Kevin Smith, 2024-02-01 A comprehensive summary of Grade 11 12 Physics Simple logical summaries with example exam questions and work through solutions The book covers the fundamentals of Grade 11 12 Physics and complements the material in any class text *Chemistry, Student Study Guide* James E. Brady, Fred Senese, 2008-01-28 The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon as it forms around particles of a metal catalyst Carbon nanotubes are a recently discovered allotrope of carbon Three other allotropes of carbon buckyballs graphite and diamond are illustrated at the left as is the molecule methane  $\text{CH}_4$  from which nanotubes and buckyballs can be made The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement **A Pronouncing Chemical Formula Speller and Contest Guide** Carl Alfred Jacobson, 1925 *Student Study Guide/Solutions Manual for Essentials of General, Organic, and Biochemistry* Denise Guinn, Rebecca Brewer, Rachel C. Lum, 2009-09-15 The Student Study Guide and Solutions Manual provides students with a combined manual designed to help them avoid common mistakes and understand key concepts After a brief review of each section s critical ideas students are taken through stepped out worked examples try it yourself examples and chapter quizzes all structured to reinforce chapter objectives and build problem solving techniques The solutions manual includes detailed solutions to all odd numbered exercises in the text *Study Guide for General Chemistry for Colleges* Herman Thompson Briscoe, 1936 **The Practice of Chemistry Study Guide & Solutions Manual** Pamela Mills, Amina El-Ashmawy, 2003-04-14 Designed to help students understand the material better and avoid common mistakes Also includes solutions and explanations to odd numbered exercises **Introductory Chemistry Study Guide** Holder, 2003-05 **Chemistry** Richard Post, Chad Snyder, Clifford C. Houk, 2020-08-17 A practical complete and easy to use guide for understanding major chemistry concepts and terms Master the fundamentals of chemistry with this fast and easy guide Chemistry is a fundamental science that touches all other sciences including biology physics electronics environmental studies astronomy and more Thousands of students have successfully used the previous editions of Chemistry Concepts and Problems A Self Teaching Guide to learn chemistry either independently as a refresher or in parallel with a college chemistry course This newly revised edition includes updates and additions to improve your success in learning chemistry This book uses an interactive self teaching method including frequent questions and study problems increasing both the speed of learning and retention Monitor your progress with self tests and master chemistry quickly This revised Third Edition provides a fresh step by step approach to learning that requires no prerequisites lets you work at your own pace and reinforces what you learn ensuring lifelong mastery Master the science

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