

Name \_\_\_\_\_ answer key \_\_\_\_\_ Date \_\_\_\_\_

# Stoichiometry: Mass – Mole Worksheet

Based on the following chemical equation, please answer the following questions!



Calculate the amount of moles of NaCl needed to produce 75 g of FeCl<sub>3</sub>.

$$\frac{9 \text{ FeCl}_3}{1 \text{ mole FeCl}_3} \times \frac{1 \text{ mole FeCl}_3}{9 \text{ FeCl}_3} \times \frac{\text{mole NaCl}}{\text{mole FeCl}_3}$$

Calculate the amount of Fe<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub> in grams produce 215 g Na<sub>2</sub>SO<sub>4</sub>.

$$\frac{9 \text{ Na}_2\text{SO}_4}{1 \text{ mole Na}_2\text{SO}_4} \times \frac{1 \text{ mole Na}_2\text{SO}_4}{9 \text{ Na}_2\text{SO}_4} \times \frac{\text{mole Fe}_2(\text{SO}_4)_3}{\text{mole Na}_2\text{SO}_4} \times \frac{9 \text{ Fe}_2(\text{SO}_4)_3}{1 \text{ mole Fe}_2(\text{SO}_4)_3}$$

Calculate the amount in grams of FeCl<sub>3</sub> produced with 95g of Fe (SO<sub>4</sub>)<sub>3</sub>

$$\frac{9 \text{ Fe}_2(\text{SO}_4)_3}{1 \text{ mole Fe}_2(\text{SO}_4)_3} \times \frac{1 \text{ mole Fe}_2(\text{SO}_4)_3}{9 \text{ Fe}_2(\text{SO}_4)_3} \times \frac{\text{mole FeCl}_3}{\text{mole Fe}(\text{SO}_4)_3} \times \frac{9 \text{ FeCl}_3}{1 \text{ mole FeCl}_3}$$

Calculate the amount of moles of Fe<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub> needed to react with 123.5 g of NaCl

$$\frac{9 \text{ NaCl}}{1 \text{ mole NaCl}} \times \frac{1 \text{ mole NaCl}}{9 \text{ NaCl}} \times \frac{\text{mole Fe}(\text{SO}_4)_3}{\text{mole NaCl}}$$

Based on the following chemical equation, please answer the following questions!



$$\frac{9 \text{ Ca}(\text{HCO}_3)_2}{1 \text{ mole Ca}(\text{HCO}_3)_2} \times \frac{1 \text{ mole Ca}(\text{HCO}_3)_2}{9 \text{ Ca}(\text{HCO}_3)_2} \times \frac{\text{mole CaCO}_3}{\text{mole Ca}(\text{HCO}_3)_2}$$

# Mass And The Mole Study Guide Answers

**Ian Pickup**



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Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual Leo J. Malone, Theodore O. Dolter, 2012-01-03 The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections assessment exercises at the end each section and relevant chapter problems at the end of each chapter A new Math Check allows quick access to the needed basic skill The first chapter now includes brief introductions to several fundamental chemical concepts and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem Every concept in the text is clearly illustrated with one or more step by step examples Making it Real essays have been updated to present timely and engaging real world applications emphasizing the relevance of the material they are learning This edition continues the end of chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter

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## **Table of Contents Mass And The Mole Study Guide Answers**

1. Understanding the eBook Mass And The Mole Study Guide Answers
  - The Rise of Digital Reading Mass And The Mole Study Guide Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Mass And The Mole Study Guide Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Mass And The Mole Study Guide Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Mass And The Mole Study Guide Answers
  - Personalized Recommendations
  - Mass And The Mole Study Guide Answers User Reviews and Ratings
  - Mass And The Mole Study Guide Answers and Bestseller Lists
5. Accessing Mass And The Mole Study Guide Answers Free and Paid eBooks
  - Mass And The Mole Study Guide Answers Public Domain eBooks
  - Mass And The Mole Study Guide Answers eBook Subscription Services
  - Mass And The Mole Study Guide Answers Budget-Friendly Options

6. Navigating Mass And The Mole Study Guide Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Mass And The Mole Study Guide Answers Compatibility with Devices
  - Mass And The Mole Study Guide Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Mass And The Mole Study Guide Answers
  - Highlighting and Note-Taking Mass And The Mole Study Guide Answers
  - Interactive Elements Mass And The Mole Study Guide Answers
8. Staying Engaged with Mass And The Mole Study Guide Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Mass And The Mole Study Guide Answers
9. Balancing eBooks and Physical Books Mass And The Mole Study Guide Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Mass And The Mole Study Guide Answers
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Mass And The Mole Study Guide Answers
  - Setting Reading Goals Mass And The Mole Study Guide Answers
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Mass And The Mole Study Guide Answers
  - Fact-Checking eBook Content of Mass And The Mole Study Guide Answers
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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