Mastering Biology Ch 04 HW

Most organic compounds contain carbon and _____. - answer Hydrogen

What is the molar ratio of methionine to glycine in the samples? - answer1.8 \times 10-3 mole of methionine per mole of glycine

How many molecules of glycine are present in 1.0 mole of glycine? - answer6.02 × 1023 molecules

For every 1.0 mole of glycine in the sample, how many molecules of methionine are present? - answer1.08 × 1021 molecules of methionine

Which amino acid is present in higher amounts than glycine in the samples? - answer Alanine

If there is 1.0 mole of glycine present in the sample, how many more molecules of alanine are present? - answer6.02 x 1022

If one mole each of CH4, NH3, H2S, and CO2 is added to 1 liter of water in a flask (1 liter of water = 55.5 moles of H2O), how many moles of hydrogen, carbon, oxygen, nitrogen, and sulfur are in the flask? Round your answers to the nearest whole number.

- answerHydrogen-120

Carbon-2 Oxygen-58 Nitrogen-1 Sulfer-1

Glycine's molecular formula is C2H5NO2. How many moles of each element would be needed to make 1.0 mole of glycine? - answer2 moles carbon, 5 moles hydrogen, 1 mole nitrogen, and 2 moles oxygen

What is the maximum number of moles of glycine (C2H5NO2) that could be made in the flask if no other molecules were made? - answer1 mole of glycine

If only serine (C3H7NO3) was made, which element would be used up first in the flask?
- answerCarbon

What is the maximum number of moles of serine that could be made in the flask if no other molecules were made? - answer2/3 (0.67) of a mole of serine

If only methionine (C5H11NO2S) was made, which element would be used up first? answerCarbon

Mastering Biology Ch 6 Answers

Jin-Ying Zhang

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