

Chemistry Spring Final Study Guide

Unit 10: Acids & Bases

Use the following information and complete the structure of Acids & Bases.

Acids	Bases
Unit 10: Acids & Bases Draw a line on the graph and indicate its dependence for each property: Temperature, Volume, and Pressure below. Label the y-axis. Label the x-axis. 1. A substance of the particles. 2. A substance of the particles. 3. A substance of the particles. 4. A substance of the particles. 5. A substance of the particles. 6. A substance of the particles. 7. A substance of the particles. 8. A substance of the particles. 9. A substance of the particles. 10. A substance of the particles.	Unit 10: Acids & Bases Draw a line on the graph and indicate its dependence for each property: Temperature, Volume, and Pressure below. Label the y-axis. Label the x-axis. 1. A substance of the particles. 2. A substance of the particles. 3. A substance of the particles. 4. A substance of the particles. 5. A substance of the particles. 6. A substance of the particles. 7. A substance of the particles. 8. A substance of the particles. 9. A substance of the particles. 10. A substance of the particles.

Unit 11: Chemical Reactions

Write the balanced chemical equation for each reaction.

1. $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$

2. $\text{CH}_4 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$

3. $\text{Fe} + \text{O}_2 \rightarrow \text{Fe}_2\text{O}_3$

4. $\text{Al} + \text{O}_2 \rightarrow \text{Al}_2\text{O}_3$

5. $\text{C}_2\text{H}_6 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$

6. $\text{C}_3\text{H}_8 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$

7. $\text{C}_4\text{H}_{10} + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$

8. $\text{C}_5\text{H}_{12} + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$

9. $\text{C}_6\text{H}_{14} + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$

10. $\text{C}_7\text{H}_{16} + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$

Unit 12: Chemical Reactions

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9. $\text{C}_6\text{H}_{14} + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$

10. $\text{C}_7\text{H}_{16} + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$

Unit 13: Chemical Reactions

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