

## ◀ Homework Chapter 6

### Exercise 6.39: Problems - The Mole Concept

#### ▼ Part A

How many carbon atoms are in a diamond (pure carbon) with a mass of 75 mg?

Express your answer using two significant figures.



$$N = 2.5 \cdot 10^{21}$$

C atoms



You have already submitted this answer. Enter a new answer.

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# Mastering Chemistry Homework Answers Answer Key

**Clemens Wendtner**



## **Mastering Chemistry Homework Answers Answer Key:**

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