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Information Theory & Coding

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Dr. J. S. Chitode

first edition : sept. 2017
fourth edition : oct. 2022



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Monica Borda



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web aug 19 2022 atoms with the same number of protons but different numbers of neutrons are called isotopes they share almost the same chemical properties but differ in mass and therefore in physical properties there are stable isotopes which do not emit radiation and there are unstable isotopes which do emit radiation

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for example carbon 14 has 8 neutrons differing from carbon 12 which has 6

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the difference in the mass is due to the difference in their number of neutrons with the same number of electrons and protons

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