

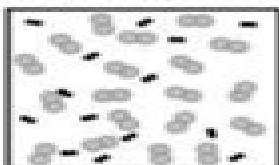
## CHAPTER 18 HOW CHEMICALS MIX

### Multiple Choice

Identify the choice that best completes the statement or answers the question.

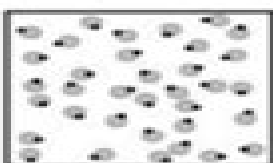
- \_\_\_\_\_ 1. A combination of two or more substances in which they no longer retain their chemical properties is called a(n) \_\_\_\_\_.  
a. heterogeneous mixture  
b. compound  
c. suspension  
d. mixture  
e. periodic trend

- \_\_\_\_\_ 2. The following image represents which kind of matter?



- a. a mixture  
b. an element  
c. a compound  
d. none of the above  
e. all of the above

- \_\_\_\_\_ 3. The following image represents which kind of matter?



- a. an element  
b. a compound  
c. a mixture  
d. none of the above  
e. all of the above

- \_\_\_\_\_ 4. Which of the following is a mixture?

- a. salt  
b. iron  
c. gold  
d. helium  
e. air

- \_\_\_\_\_ 5. How does a suspension differ from a solution?

- a. A suspension is a heterogeneous mixture consisting of different phases whereas a solution is a homogeneous mixture consisting of a single phase.  
b. The difference between a suspension and a solution can only be determined by chemical means.  
c. Although a solution and suspension are both homogeneous mixtures, only the components of a suspension will separate by spinning the mixture in a centrifuge.  
d. A suspension is a heterogeneous mixture whose components can be separated by simple filtration. A solution is a homogeneous mixture which cannot be separated by simple

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