

ACID-BASE STUDY GUIDE

- Normal Plasma pH 7.35 to 7.45 (hydrogen (H⁺) concentration)
 - Higher this goes, higher the mortality rate
 - Measure the acidity or alkalinity of blood
 - Homeostatic mechanisms keep pH normal
 - Buffer systems, kidneys & lungs
 - Greater H⁺ concentration = lower pH & more acidic solute
 - Lower H⁺ concentration = higher pH & more alkaline solute
- Major ECF buffer system
 - Bicarbonate-carbonic acid system
 - $\text{CO}_2 + \text{H}_2\text{O} \leftrightarrow \text{H}_2\text{CO}_3 \leftrightarrow \text{H}^+ \text{ and } \text{HCO}_3^-$
 - Kidneys regulate bicarbonate in the ECF
 - Lungs (controlled by medulla) regulate CO₂, thus the carbonic acid in the ECF
 - Lungs are first responders
- Other buffer systems
 - ECF: inorganic phosphates and plasma proteins
 - ICF: proteins, organic, inorganic phosphates
 - Hemoglobin

METABOLIC ACIDOSIS

pH lower than 7.35

- Can occur by a gain of H⁺ or loss of bicarbonate ions (<22 mm Hg)
- Causes
 - Direct loss of bicarbonate (diarrhea, administration of IV nutrition w/o bicarb or bicarb-producing solutes, diuretic use, early renal insufficiency, excess chloride, lower intestinal fistulas, ureterostomies)
 - Most common due to kidney injury
 - Lactic acidosis (higher it is, the worse sepsis is), salicylate poisoning, renal failure, methanol, ethylene, or propylene toxicity, DKA, ketoacidosis due to starvation
 - Systemic issue: diabetes
- Clinical Manifestations
 - Headache, confusion, drowsiness, increased RR/depth, low BP, nausea, vomiting, cold/clammy skin, arrhythmias, shock, peripheral vasodilation, decreased cardiac output
- Medical Management
 - First have to correct underlying issue

Key Acid Base Study Guide

**Mariann M. Harding, Collin Bowman-
Woodall, Jeffrey Kwong, Dottie
Roberts, Debra Hagler, Courtney
Reinisch**

Key Acid Base Study Guide:

Decoding **Key Acid Base Study Guide**: Revealing the Captivating Potential of Verbal Expression

In an era characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its capability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Key Acid Base Study Guide**," a mesmerizing literary creation penned by a celebrated wordsmith, readers attempt an enlightening odyssey, unraveling the intricate significance of language and its enduring affect our lives. In this appraisal, we shall explore the book is central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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