

**Double-Replacement Precipitation Lab**

**Introduction:** Double-Replacement reactions occur when two ionic solutions mix to form an insoluble solid. If all cations and anions are soluble then no reaction will occur. During the lab procedure, several ionic solutions will be mixed to determine if a precipitate reaction will occur. Those reactions that do not form a precipitate will be marked for No Reaction (NR). Students will practice writing and balancing double-replacement reactions.

**Pre-Lab questions:**

1. How would you describe a double replacement reaction to a friend who is struggling with reaction types?
2. What is a precipitate?

**Materials:**

- 0.2M sodium bromide, NaBr
- 0.2M sodium carbonate,  $\text{Na}_2\text{CO}_3$
- 0.2M sodium phosphate,  $\text{Na}_3\text{PO}_4$
- 0.2M sodium sulfate,  $\text{Na}_2\text{SO}_4$
- 0.2M sodium hydroxide, NaOH
- Transparency paper
- 0.2M lithium chloride, LiCl
- 0.2M copper(II) chloride,  $\text{CuCl}_2$
- 0.2M zinc chloride,  $\text{ZnCl}_2$
- 0.2M strontium chloride,  $\text{SrCl}_2$
- 0.2M barium chloride  $\text{BaCl}_2$
- Background paper Template

**Safety:** You MUST wear goggles during this lab activity

**Procedure:**

1. Place the overhead sheet over the background paper.
2. Using the Data Table as a guide, mix every possible pair of solutions in a separate square of the background paper. **Use 1 or 2 drops of each solution.** Do NOT contaminate the individual pipets with different solutions.
3. Mix the solutions with a rinsed toothpick if you are uncertain if a precipitate formed. Rinse the toothpick after EACH mixing – be careful of which end of the rod you use so that you do not touch the chemicals or contaminate your samples!
4. Observe each mixture carefully for signs of a precipitate. (Some precipitates are light in color and hard to see so you may want to slide the transparency over the black lab top on occasion.) Record your observations (presence or absence of a precipitate, and other characteristics such as color, texture, etc.) in the Data Table. If nothing happens, write NR for no reaction.
5. Rinse and wipe the transparency sheet into the sink. Dry the sheet.

**Observations:** Write the observations you made from combining these materials in the spaces in the table below. Note any color changes or precipitate (cloudiness/solid). If no change is observed write NR for no reaction.

	NaBr	$\text{Na}_2\text{CO}_3$	$\text{Na}_3\text{PO}_4$	$\text{Na}_2\text{SO}_4$	NaOH
LiCl					
$\text{CuCl}_2$					
$\text{ZnCl}_2$					
$\text{SrCl}_2$					
$\text{BaCl}_2$					

# Lab 27 Double Replacement Reactions

**AN Whitehead**



## Lab 27 Double Replacement Reactions:

*CHEMISTRY EXPERIMENTS* James Signorelli, 2014-09-19 Gifted and talented students and any student interested in pursuing a science major in college needs a rigorous program to prepare them while they are still in high school This book utilizes a format where the application of several disciplines science math and language arts principles are mandated Each lab concludes with either an essay or a detailed analysis of what happened and why it happened This format is based on the expectations of joining a university program or becoming an industrial science professional The ideal student lab report would be written in a lab research notebook and then the essay or final analysis is done on a word processor to allow for repeat editing and corrections The research notebook has all graph pages a title section and a place for the students and their assistants to sign and witness that exercise The basic mechanics of the lab report title purpose procedure diagrams data table math and calculations observations and graphs are handwritten into the book The conclusion is done on a word processor MS Word which allows the instructor to guide the student in writing and editing a complete essay using the MLA format When the final copy is completed the essay is printed and inserted into the lab notebook for grading At the end of the term the student has all their labs in one place for future reference These lab notebooks can be obtained for as little as 3 00 per book This is money well spent In our district the Board of Education buys the books for each student The BOE sees these books as expendable but necessary materials for all science and engineering instruction Chemistry Eugene LeMay, Jr., Herbert Beall, Karen M. Robblee, Douglas C. Brower, Prentice-Hall Staff, 2002-02 *Top Shelf* Brian Pressley, 2003 Covers chemical formulas and equations chemical reactions structure of atoms the gas laws and more Presents hands on activities as catalysts to fuel student imagination CHEMICAL REACTIONS NARAYAN CHANGDER, 2024-04-08 Note Anyone can request the PDF version of this practice set workbook by emailing me at cbsenet4u gmail com You can also get full PDF books in quiz format on our youtube channel <https://www.youtube.com/smartquiziz> I will send you a PDF version of this workbook This book has been designed for candidates preparing for various competitive examinations It contains many objective questions specifically designed for different exams Answer keys are provided at the end of each page It will undoubtedly serve as the best preparation material for aspirants This book is an engaging quiz eBook for all and offers something for everyone This book will satisfy the curiosity of most students while also challenging their trivia skills and introducing them to new information Use this invaluable book to test your subject matter expertise Multiple choice exams are a common assessment method that all prospective candidates must be familiar with in today's academic environment Although the majority of students are accustomed to this MCQ format many are not well versed in it To achieve success in MCQ tests quizzes and trivia challenges one requires test taking techniques and skills in addition to subject knowledge It also provides you with the skills and information you need to achieve a good score in challenging tests or competitive examinations Whether you have studied the subject on your own read for pleasure or completed coursework it will assess

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This book delves into Lab 27 Double Replacement Reactions. Lab 27 Double Replacement Reactions is an essential topic that must be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Lab 27 Double Replacement Reactions, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:

- Chapter 1: Introduction to Lab 27 Double Replacement Reactions
- Chapter 2: Essential Elements of Lab 27 Double Replacement Reactions
- Chapter 3: Lab 27 Double Replacement Reactions in Everyday Life
- Chapter 4: Lab 27 Double Replacement Reactions in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, the author will provide an overview of Lab 27 Double Replacement Reactions. This chapter will explore what Lab 27 Double Replacement Reactions is, why Lab 27 Double Replacement Reactions is vital, and how to effectively learn about Lab 27 Double Replacement Reactions.

3. In chapter 2, the author will delve into the foundational concepts of Lab 27 Double Replacement Reactions. This chapter will elucidate the essential principles that need to be understood to grasp Lab 27 Double Replacement Reactions in its entirety.

4. In chapter 3, this book will examine the practical applications of Lab 27 Double Replacement Reactions in daily life. The third chapter will showcase real-world examples of how Lab 27 Double Replacement Reactions can be effectively utilized in everyday scenarios.

5. In chapter 4, this book will scrutinize the relevance of Lab 27 Double Replacement Reactions in specific contexts. The fourth chapter will explore how Lab 27 Double Replacement Reactions is applied in specialized fields, such as education, business, and technology.

6. In chapter 5, the author will draw a conclusion about Lab 27 Double Replacement Reactions. This chapter will summarize the key points that have been discussed throughout the book.

The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Lab 27 Double Replacement Reactions.

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