Name _____ Skills Practice Lab Class.

OBSERVATION

п

Date

Evidence for Chemical Change

One way of knowing that a chemical change has occurred is to observe that the properties of the products are different from those of the reactants. A new product can then become a reactant in another chemical reaction. In this experiment, you will observe a sequence of changes that occur when a solution that begins as copper(II) nitrate is subjected to a series of different reactions. All of the reactions will take place in the same test tube. At each step, you will look for evidence that a new substance has formed as a result of a chemical change. You will also observe energy changes and relate them to chemical reactions.

OBJECTIVES

Observe evidence that a chemical change has taken place.

Infer from observations that a new substance has been formed.

Identify and record observations that show energy is involved in chemical change.

Observe the color and solubility of some substances.

Describe reactions by writing word equations.

MATERIALS

- aluminum wire, 12 cm
- beaker, 100 mL
- Bunsen burner and related equipment
- copper(II) nitrate, 0.75 M
- glass stirring rod
- gloves
- HCl, 3.0 M
- iron ring.

- lab apron
- lab marker
- NaOH, 3.0 M
- ring stand
- maler
- safety goggles
- test tube, 13 mm × 100 mm
- test-tube rack
- wire gauze, ceramic-centered

Always wear safety goggles and a lab apron to protect your eyes and clothing. If you get a chemical in your eyes, immediately flush the chemical out at the eyewash station while calling to your teacher. Know the location of the emergency lab shower and the eyewash station and the procedures for using them.

Do not touch any chemicals. If you get a chemical on your skin or clothing, wash the chemical off at the sink while calling to your teacher. Make sure you carefully read the labels and follow the precautions on all containers of chemicals that you use. If there are no precautions stated on the label, ask your teacher what precautions to follow. Do not taste any chemicals or items used in the laboratory. Never return leftover chemicals to their original containers; take only small amounts to avoid wasting supplies.

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Lab Evidence For Chemical Change Answers

Michael J. Padilla

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Arena, Cary Willard, 2023 Foundations of College Chemistry 16th edition presents chemistry as a modern vital subject and is designed to make introductory chemistry accessible to all beginning students. It is intended for students who have never taken a chemistry course or those who had a significant interruption in their studies but plan to continue with the general chemistry sequence. The central focus is to make chemistry interesting and understandable and teach students the problem solving skills they will need. This International Adaptation offers new and updated content with improved presentation of all course material. It builds on the strengths of previous editions including clear explanations and step by step problem solving. The material emphasizes real world applications of chemistry as the authors develop the principles that form the foundation for the further study of chemistry. There is new and expanded coverage of polarizing power and polarizability Fajans rules collision number and mean free path abnormal molecular masses and van t Hoff factor and applications of radioactivity

Lab Experiments for Modern Chemistry Tzimopoulo,1990 The Life Cycle of Everyday Stuff Mike Reeske, Shirley Watt Ireton, National Science Teachers Association, 2001 In the context of life cycles these units use central science concepts to explore the energy raw materials and waste issues that are the history of any manufactured product As students consider the trade offs made at each step they will learn to recognize the decisions made to balance economic developmental and environmental needs Introduction to General, Organic, and Biochemistry Morris Hein, Scott Pattison, Susan Arena, Leo R. Best, 2014-01-15 The most comprehensive book available on the subject Introduction to General Organic and Biochemistry 11th Edition continues its tradition of fostering the development of problem solving skills featuring numerous examples and coverage of current applications Skillfully anticipating areas of difficulty and pacing the material accordingly this readable work provides clear and logical explanations of chemical concepts as well as the right mix of general chemistry organic chemistry and biochemistry. An emphasis on real world topics lets readers clearly see how the chemistry will apply to their Predict, Observe, Explain John Haysom, Michael Bowen, 2010 John Haysom and Michael Bowen provide middle and career high school science teachers with more than 100 student activities to help the students develop their understanding of scientific concepts The powerful Predict Observe Explain POE strategy field tested by hundreds of teachers is designed to foster student inquiry and challenge existing conceptions that students bring to the classroom **Modern Chemistry** Nicholas D. Tzimopoulos, 1990 Foundations of College Chemistry Morris Hein, Susan Arena, 2013-01-01 Learning the

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<u>Science Explorer: Chemical Interactions</u> Pearson/Prentice Hall, Michael J. Padilla, 2004-10 **CSE Report**, 2006 Prentice Hall Science Explorer Michael J. Padilla, 2002 **Scientific and Technical Aerospace Reports**, 1992

Fundamentals of Microbiology Jeffrey C. Pommerville, 2014-12 Ideal for health science and nursing students Fundamentals of Microbiology Body Systems Edition Third Edition retains the engaging student friendly style and active learning approach for which award winning author and educator Jeffrey Pommerville is known Highly suitable for non science majors the fully revised and updated third edition of this bestselling text contains new pedagogical elements and an established learning design format that improves comprehension and retention and makes learning more enjoyable Unlike other texts in the field Fundamentals of Microbiology Body Systems Edition takes a global perspective on microbiology and infectious disease and supports students in self evaluation and concept absorption Furthermore it includes real life examples to help students understand the significance of a concept and its application in today s world whether to their local community or beyond New information pertinent to nursing and health sciences has been added while many figures and tables have been updated revised and or reorganized for clarity Comprehensive yet accessible the Third Edition is an essential text for non science majors in health science and nursing programs taking an introductory microbiology course Educational Film & Video Locator of the Consortium of College and University Media Provided by publisher **Centers and R.R. Bowker** Consortium of College and University Media Centers,1990 **Defining Multiple Chemical Sensitivity** Bonnye L. Matthews, 2015-08-13 The number of people suffering from Multiple Chemical Sensitivity MCS is greater than the number suffering from AIDS yet the general public and the medical community know very little about the disease More and more people are disabled daily despite the fact that the condition does not have to occur In Part One of this work experts review the research into the disease along with treatment strategies Part Two examines the legal recourses available to MCS sufferers such as workers compensation claims and product liability suits How the medical community has often worked against MCS sufferers is the focus of Part Three demonstrating that medical opposition to the disease is unfounded The editor an MCS sufferer details her own case in Part Four

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