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Honda Xr650l Repair Manual On Carb Book Review: Unveiling the Power of Words

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web using algebra the moles was founded to be 0 00352 moles of butane in order to find the molarity the mass of the butane should be divided by the moles after further calculations the molarity of the butane gas was calculated to be 56 81 grams moles this lab is done with tons of sources of error

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web this data shows that there was a gain the mass of the lighter which shows that there was a certain amount of butane 0 38 grams 0 02 grams gas that was successfully collected during the gas collection there is a mild influence of barometric pressure during both trials as shown above in the raw data

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gas law the number of moles of butane gas and then the molar mass of butane can be calculated your experimental value for molar mass will be compared to the known molar mass of butane and percent error calculated

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