



# Word Interactive Tutorial

## How to Create a Word Document

# Make A Word Document

**Jianjun Gao**



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web write down the equilibrium equation you are investigating using the information provided by the introduction within the virtual lab activity equilibrium reaction  $CoH_2O_6 + 2ClCOCl_2 \rightleftharpoons 6H_2O$  select each control test tube to compare the concentrations of

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web equilibrium moles  $hscn$   $2.00 \times 10^{-5} \text{ mol}$   $3.00 \times 10^{-6} \text{ mol}$   $1.70 \times 10^{-5} \text{ mol}$   $hscn$  knowing the number of moles of  $Fe^{3+}$  and  $hscn$  present in the equilibrium mixture and the volume of the mixture we can easily find the concentrations of those two species  $mol\ Fe^{3+}$   $1.70 \times 10^{-5} \text{ mol}$   $Fe^{3+}$   $8.50 \times 10^{-4} \text{ M}$

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