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**Victor M. Corman**



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web solution we use the relation  $n = \frac{m}{M} \times \frac{t}{T}$  where  $n$  is the number of half-lives,  $m$  is the mass of the sample,  $M$  is the molar mass,  $t$  is the time, and  $T$  is the half-life. For example, if a sample of 100 g of a substance with a half-life of 4 days decays to a mass of 30 mg in 16 days, we can calculate the initial mass of the sample. When we have  $n$  and  $t$ , rearranging the equation used in example 2 we obtain

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web solution 20.036556 half lives 0.0213 the decimal fraction remaining after 5.56 half lives  $6.02 \times 10^{-2} \times 0.0213^{1.28} \times 10$  problem 3 os-182 has a half life of 21.5 hours how many grams of a 10.0 gram sample would have decayed after exactly three half lives solution 0.125 the amount remaining after 3 half lives

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