

# SECTION 1 Set-Up & First Brew

**IMPORTANT:** You cannot use your Keurig Brewer until it has been set-up for its first use as set forth below.

**NOTE:** If Brewer has been exposed to temperatures below freezing, allow Brewer at least two hours to reach room temperature before brewing. A frozen or extremely cold brewer will not operate.



Fig. 1 Setting Up Your Brewer

1. Unwrap cord and plug the Brewer into a grounded outlet. Remove all pieces of packing tape and clear plastic sticker from LCD display.
2. Place a large mug (10 oz. min) on the Drip Tray Plate.
3. Remove the Water Reservoir from Brewer by first removing Lid from Reservoir (Fig. 1), then lifting Reservoir straight up and away from Brewer (may be tight), (Fig. 2). Rinse with fresh water.

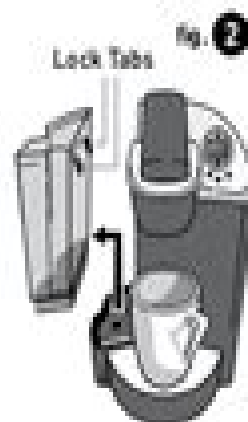


Fig. 2

4. Fill Water Reservoir with bottled or filtered water so that the water level is just below the top edge of the Water Reservoir.
5. Place Water Reservoir base into the left side of the Brewer and be sure the Water Reservoir Lock Tabs engage with the Brewer. Replace Lid on Water Reservoir.
6. Press the Power Button found directly above the LCD Control Center (Fig. 3).



Fig. 3

7. The LCD Control Center will come on and indicate "NOT READY" and the Small Mug Button will also flash. **NOTE:** A K-Cup should not be used for this Sequence.
8. Press the Small Mug Button (Fig. 3). **NOTE:** If the Water Reservoir is empty or not in place on the Brewer, the LCD Control Center will also indicate "ADD WATER." None of the buttons will be activated until water is added to the Water Reservoir.

# Keurig B60 Coffee Maker Manual

**J Rink**



## **Keurig B60 Coffee Maker Manual:**

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web this is like a one dimensional system whose mechanical energy  $e$  is a constant and whose potential energy with respect

to zero energy at zero displacement from the spring's unstretched length  $x = 0$  is  $U = \frac{1}{2} kx^2$  figure 8.5.2 a a glider between springs on an air track is an example of a horizontal mass spring system

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web 1 if the  $pe$  of the reactants is 430 kJ and the  $pe$  of the products is 190 kJ what is the  $\Delta H$  2 if the  $pe$  of the reactants is 430 kJ and the activation energy is 220 how much total energy will be needed to be added for the reaction to proceed transcribed image text potential energy diagram answer the questions using the graph

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web figure 5.5.1 a potential energy diagram shows the total potential energy of a reacting system as the reaction proceeds a in an endothermic reaction the energy of the products is greater than the energy of the reactants and  $\Delta H$  is positive b in an exothermic reaction the energy of the products is lower than the energy of the

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