# Section 15.1 What are solutions?

in your fortbook, read about the characteristics of solutions.

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Study Guide for Content Mastery Answer Key

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Because the sodium ion is positively charged, it attracts the negatively charged 76 chloride ion is negatively charged, it attracts the positively charged portion charged portion of the water molecule (the hydrogen atoms). Because the portion of the water molecule (the oxygen atom) and repels the positively Explain the receivation of the water molecules around the reclaim tons and obtainfollows. the water molecule and repels the negatively charged portion.

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The attraction between the water molecules and the sodium and chloride lons is strength of attaction between the water molecules and the ions is what causes greater than the attaction between the sodium and chloride ions. The greater the solvation process to extra

stiring or shaking the solution, breaking the solute into smaller pieces, and G. List there was that the rate of solvation may be increased. heating the solvent

is your tenthook, read about heat of solution, solubility, and factors that affect solubility

For each statement below, write iron or John

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