

## THE CONSERVATION OF LINEAR MOMENTUM

### Introduction

In this experiment you will test the validity of the Law of Conservation of Linear Momentum in one dimension utilizing elastic and inelastic collisions on an air track.

### Apparatus

Computer with Logger Pro software

Vernier Lab Pro box

Pasco Air Track

Air supply

Air Track Accessory kit - bumpers for the elastic collision

Air Track Gliders (2)

Vernier Photogate (2)

Ring Stands (2)

Right angle clamps (2) or integral photogate clamps

Laboratory Balance

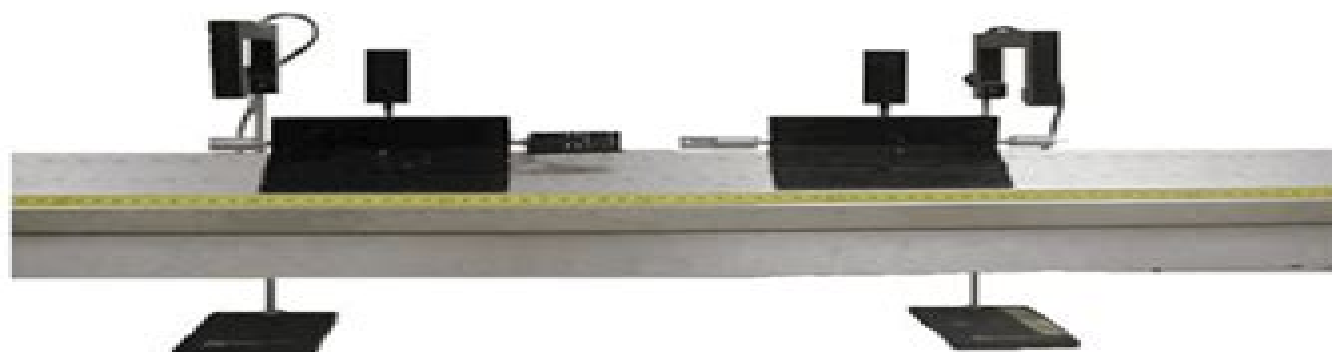


Figure 0: Elastic Near Collision

### Theory

If two objects collide, and are subject to no net external forces, then it can be shown by application of Newton's 2nd and 3rd Laws that the total linear momentum of the system of masses will not be altered by the collision. The linear momentum of an object of mass  $m_1$  and velocity  $v_1$  is given by  $p_1 = m_1 v_1$ . In a system consisting of two objects of momentum  $p_1$  and  $p_2$ , the total linear momentum is the vector sum of their individual momenta:

$$\mathbf{p}_1 + \mathbf{p}_2 = m_1 \mathbf{v}_1 + m_2 \mathbf{v}_2$$

The total linear momentum *before* collision is  $m_1 \mathbf{v}_1 + m_2 \mathbf{v}_2$

# Linear Momentum Lab Manual Solutions

**Daniela Niemeyer**



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**A Brief Introduction to Fluid Mechanics** Donald F. Young, Bruce R. Munson, Theodore H. Okiishi, Wade W. Huebsch, 2010-12-21 A Brief Introduction to Fluid Mechanics 5th Edition is designed to cover the standard topics in a basic fluid mechanics course in a streamlined manner that meets the learning needs of today's student better than the dense encyclopedic manner of traditional texts This approach helps students connect the math and theory to the physical world and practical applications and apply these connections to solving problems The text lucidly presents basic analysis techniques and addresses practical concerns and applications such as pipe flow open channel flow flow measurement and drag and lift It offers a strong visual approach with photos illustrations and videos included in the text examples and homework problems to emphasize the practical application of fluid mechanics principles      **Radioactive Waste Management**, 1988      Scientific and Technical Aerospace Reports, 1995      **NBS Special Publication**, 1973      **Government Reports Announcements**, 1974      The Publishers' Trade List Annual, 1978      Energy Research Abstracts, 1993      *Use of Services for Family Planning and Infertility, United States* Gerry E. Hendershot, Karl E. Bauman, 1988      Nuclear Science Abstracts, 1966

Hydraulic Research in the United States and Canada United States. National Bureau of Standards, 1972      **Hydraulic Research in the United States and Canada, 1972** United States. National Bureau of Standards, 1974      **Report on the High Speed Ground Transportation Act of 1965**, 1971      **Report on the High Speed Ground Transportation Act** United States. Department of Transportation,      **ERDA Energy Research Abstracts**, 1983      *R.R. Bowker's Software for Schools*, 1987      *Technical Publications Announcements with Indexes* United States. National Aeronautics and Space Administration, 1962      **Guide to Annual Subject Index for Technical Publications Announcements, Apr.-Dec. 1962** United States. National Aeronautics and Space Administration, 1962

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 $2 \text{ CO}_2 \text{ g} \rightarrow \text{C}_2\text{H}_2 \text{ g} + \frac{5}{2} \text{ O}_2 \text{ g}$   $\Delta h = 94.5 \text{ kJ}$   
 $\text{H}_2 \text{O g} \rightarrow \frac{1}{2} \text{ O}_2 \text{ g} + \text{H}_2 \text{ g}$   $\Delta h = 71.2 \text{ kJ}$   
 $\text{C}_2\text{H}_6 \text{ g} \rightarrow \frac{7}{2} \text{ O}_2 \text{ g} + 2 \text{ CO}_2 \text{ g} + 3 \text{ H}_2 \text{O g}$   $\Delta h = 283 \text{ kJ}$

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 $\text{C}_2\text{H}_4 \text{ g} + 3 \text{ O}_2 \text{ g} \rightarrow 2 \text{ CO}_2 \text{ g} + 2 \text{ H}_2\text{O l}$   $\Delta h = 1411 \text{ kJ/mole}$   
 $\text{C}_2\text{H}_6 \text{ g} + 7/2 \text{ O}_2 \text{ g} \rightarrow 2 \text{ CO}_2 \text{ g} + 3 \text{ H}_2\text{O l}$   $\Delta h = 1560 \text{ kJ/mole}$   
 $\text{H}_2 \text{ g} + 1/2 \text{ O}_2 \text{ g} \rightarrow \text{H}_2\text{O l}$   $\Delta h = 285 \text{ kJ/mole}$

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 $\text{C}_2\text{H}_4 \text{ g} + 2 \text{ NO}_2 \text{ g} \rightarrow \text{C}_2\text{H}_6 \text{ g} + 2 \text{ O}_2 \text{ g} + \text{N}_2\text{O}_5 \text{ g}$

given the following information

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web hess s law labs by austin lee alayna baron lily zmachinski introduction in order to calculate the enthalpy change for the combustion of magnesium oxide  $\text{Mg} + \frac{1}{2}\text{O}_2 \rightarrow \text{MgO}$  we used

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web mar 13 2023 be able to use hess s law to calculate reaction enthalpies first law of thermodynamics energy is the ability to do work or transfer heat work is the transfer of energy from one body to another in a sense work is energy in the process of transfer

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web thus for reaction 9 2 in which 1 mole of  $\text{H}_2\text{SO}_4$  and 1 mole of  $\text{OH}^-$  combine to form 2 moles of  $\text{H}_2\text{O}$   $\Delta H = -55.8 \text{ kJ}$  as shown note that the enthalpy change  $\Delta H_{\text{products}} - \Delta H_{\text{reactants}}$  is positive if heat is absorbed that is if  $\Delta H_{\text{products}} > \Delta H_{\text{reactants}}$

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web objective the purpose of the lab is to combine two reactants in the calorimeter and measure the heats of reaction in solution eventually allowing an individual to prove or disprove hess s law introduction thermodynamics is the study of energy exchanged between the system and the surroundings

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web 3 12 hess s law in this experiment you will measure the amount of heat released in these three related exothermic reactions  $\text{NaOH}(\text{s}) + \text{Na}(\text{aq}) \rightarrow \text{OH}(\text{aq})$   $\Delta H = -1.2 \text{ kJ}$   $\text{NaOH}(\text{s}) + \text{H}(\text{aq}) \rightarrow \text{Cl}(\text{aq})$   $\Delta H = -2.0 \text{ kJ}$   $\text{Na}(\text{aq}) + \text{Cl}(\text{aq}) \rightarrow \text{H}_2(\text{g})$   $\Delta H = -2.3 \text{ kJ}$

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