

5 reaction TYPES Lab

afety:

- Students must wear goggles and tie back their hair
- Students should wear gloves
- Students should not have dangling or loose clothing that could catch fire
- Exercise caution around the flame
- Avoid completing the experiment in a room with smoke or

Equipment:

- Balance
- Graduated Pipets, 3
- Bunsen Burner
- Butane Safety Lighter
- Spatula (for weighing)
- Test Tube Clamps
- Test Tube Rack
- Test Tubes, 3
- Crucible Tongs
- Watch Glass, 2
- Wood Splint, 1

Chemicals:

- Calcium Chloride Solution, 0.5M, CaCl_2 , 3mL
- Copper Sulfate Solution, 0.5M, CuSO_4 , 5mL
- Magnesium Strips, one small, and one 2in
- Ethyl Alcohol, $\text{CH}_3\text{CH}_2\text{OH}$, 1mL
- Sodium Bicarbonate, NaHCO_3 , 1g
- Sodium Carbonate Solution, 0.5M, Na_2CO_3 , 3mL

Reaction 4:

Caution: Copper (II) Sulfate will irritate the skin

1. Place a test tube in the test tube rack
2. Obtain a short piece (around 5 cm) long of magnesium
3. Add 2mL of 0.5 M Copper (II) Sulfate solution

a. Initial Observations:

b. Observation after 5 minutes:

Caution: perform this every time when adding chemical solutions. It is flammable

1. Weigh out and add 1g of sodium bicarbonate into a test tube
- a. Observations before heating:

2. Light the Bunsen burner
3. Hold the test tube with your test tube clamp
4. Place the test tube over the Bunsen burner for about 1 minute
- a. Observations after heating:

5. Hold the test tube back over the flame
6. Move your test tube over the side of a wood splint and hold it over the mouth of the test tube
- a. Initial Observations? Record here

7. Turn off the burner

INCLUDES 5 MINI LABS AND A BALANCING PROBLEM FOR EACH LAB!

Grade 10 Types Of Reactions Lab

SA Dillow



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