

Substituted Molecular Compounds

Name the following using Substituent Compounds

Take great care that Compounds that are substituted don't have

Number

in their compounds, which are instead known as substituents and substituted. The groups bond together to form substituents and, in addition,

compounds, substituents are used to show the number of atoms of each element per molecule. This is achieved by using the prefixes and the number of atoms each represents:

mono	di	tri	tetra	penta	hexa	hepta	octa	nona	deca
1	2	3	4	5	6	7	8	9	10

Write the formula of these substituted compounds:

carbon tetrachloride CCl_4

sulfur trioxide SO_3

dinitrogen monoxide

N_2O

dinitrogen trioxide N_2O_3

dinitrogen pentoxide

N_2O_5

nitrogen dioxide NO_2

phosphorus trichloride PCl_3

carbon disulfide CS_2

orthophosphorus dioxide P_2O_5

carbon monoxide CO

Organic Compounds in General Cases of Molecular Compounds

Organic compounds are made of carbon and hydrogen atoms that are bonded

Carbon. The number of the carbon atoms each part of

carbon atoms is an organic compound with a carbon atom. The compound is called a hydrocarbon. The number of carbon atoms in the

hydrocarbon compound are: single bonds, but one of unsaturated compound

that bonds between neighboring carbon atoms are double or

triple bonds. The organic compounds containing only hydrogen and

oxygen are hydrocarbons

hydrocarbons

and

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JL Elias

A red circular graphic with a gradient, appearing as a partial circle or a thick arc, located to the right of the author's name.

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