

THE GOSPEL FOR  
AN AGE OF DOUBT

Henry Van Dyke

# Gospel Age Doubt Classic Reprint

**Marcel A. Müller**



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$$\text{ce zn cucl}_2 \rightarrow \text{zncl}_2 \text{ cu}$$
  
 how many moles of ce zncl<sub>2</sub> will be produced from 23.0 g of ce zn assuming ce cucl<sub>2</sub> is available in excess

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$$\text{br } 1 \text{ n}_2 + 3 \text{ f}_2 \rightarrow 2 \text{ nf}_3$$
  

$$\text{br } 2 \text{ c}_6\text{h}_{10} + 17 \text{ o}_2 \rightarrow 12 \text{ co}_2 + 10 \text{ h}_2\text{o}$$
  

$$\text{br } 3 \text{ hbr} + 1 \text{ khco}_3 + 1 \text{ h}_2\text{o} \rightarrow 1 \text{ kbr} + 1 \text{ co}_2 + \text{br } 4 \text{ ga} + 2 \text{ br}_3 + 3 \text{ na}_2\text{so}_3 + 1 \text{ ga}_2\text{so}_3 + 3 \text{ nabr}$$
  

$$\text{br } 5 \text{ sno} + 2 \text{ nf}_3 + 3 \text{ snf}_2 + 1 \text{ n}_2\text{o}_3 \rightarrow \text{br using the following equation}$$
  

$$\text{br } 2 \text{ naoh} + \text{h}_2\text{so}_4 \rightarrow 2 \text{ h}_2\text{o} + \text{na}_2\text{so}_4$$

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$$2 \text{ c}_6\text{h}_{10} + 17 \text{ o}_2 \rightarrow 12 \text{ co}_2 + 10 \text{ h}_2\text{o}$$
  

$$\text{hbr} + \text{khco}_3 + \text{h}_2\text{o} \rightarrow \text{kbr} + \text{co}_2 + \text{ga} + 2 \text{ br}_3 + 3 \text{ na}_2\text{so}_3 + 1 \text{ ga}_2\text{so}_3 + 3 \text{ nabr} + \text{sno} + \text{nf}_3 + \text{snf}_2$$

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 answer ce sio<sub>2</sub> + 3c → sic + 2co

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$$\text{hbr} + \text{khco}_3 + \text{h}_2\text{o} \rightarrow \text{kbr} + \text{ga} + \text{br}_3 + \text{na}_2\text{so}_3 + \text{ga}_2\text{so}_3 + \text{sno} + \text{nf}_3 + \text{snf}_2 + \text{n}_2\text{o} + \text{co}_2 + \text{nabr}$$
  
 solve the following stoichiometry grams grams problems 6

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