

Guided Reading

- In your science notebook, neatly answer 18 questions as you read the CHEM TALK for Ch1 Section 1.

ChemTalk

Open Words
element, any material that cannot be broken down into simpler materials.

THE STRUCTURE OF MATTER

In this investigation you used electricity to decompose water into two gases. You learned that the gases were different because they reacted differently to the burning and glowing splints. Since water is referred to as H_2O , a first guess would be that hydrogen (H) and oxygen (O) were produced in the experiment. The test for hydrogen is a small explosion when exposed to a burning splint. The test for oxygen is the re-ignition of a glowing splint. If you look back on the results of the experiment, you find that the volume of the hydrogen gas was twice as much as the oxygen gas. There was twice as much hydrogen as oxygen, that's where the "2" comes from in the chemical formula H_2O .

Hydrogen and oxygen are elements. An **element** is any material that cannot be broken down into simpler materials by chemical means. You are probably familiar with many elements like hydrogen, oxygen,

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