

CHEMICAL BONDS

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READING COMPREHENSION & QUESTIONS

SCIENCE REAL

Types of Chemical Bonds Reading Comprehension

What is the difference between a molecule and a compound?

What of the substances shown in the diagram above are molecules?

What of the substances shown in the diagram above are compounds?

Chemical Bonds and Electronegativity

Electronegativity is an atom's ability to attract electrons. Elements with low electronegativity have outer shells that are almost empty and tend to gain or lose electrons easily. Elements with high electronegativity have outer shells that are nearly full and tend to hold on to their electrons. Ionic bonds form when the electronegativity of two atoms differs significantly, and covalent bonds form when electronegativity of the atoms are more similar. Typically, the greater the difference in electronegativity between two elements, the more ionic the bond. Elements with similar electronegativity form covalent bonds.

Electronegativity increases from bottom left to top right in the periodic table.

Diagram A: Which atom has a higher electronegativity?

Diagram B: Which set is a cation and which set is an anion?

Diagram C: Which set of compounds are ionic?

GUIDED READING ACTIVITY

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