

Acids, Bases, Solutions Study Guide

4 pages + 4-page Answer Key

chemistry

Physical Science

SOLUTIONS, ACIDS, AND BASES STUDY GUIDE

What is a heterogeneous mixture? Provide two examples.

Describe the characteristics of a saturated solution and an unsaturated solution.

Define the term solvent and provide an example.

What is solubility, and how is it measured?

Describe the difference between concentrated and dilute.

What are the properties of acids?

What are the properties of bases?

What is the pH scale?

What is the difference between strong acids and weak acids?

What is the difference between strong bases and weak bases?

What is the difference between a strong acid and a weak acid?

What are the properties of bases?

What is the significance of the pH scale?

How does the concentration of an acid affect its pH?

What happens during neutralization?

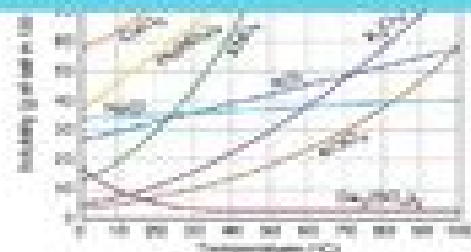
What happens when strong acids and bases are mixed?

What happens when a strong acid is diluted?

What happens when a strong base is diluted?

What happens when a strong acid reacts with a metal?

What happens when a strong base reacts with a metal?



20. Describe how you would use the graph to determine the amount of Na_2SO_4 that can be dissolved in 100 grams of water at 40°C .

22. Which salt has the highest solubility at 40°C ?



ANSWER KEY INCLUDED

Holt Chemistry Acids And Bases Study Guide

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Buckner, Gale R. Kirschner**



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