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Yersin, 2003-07-01 There exists a large literature on the spectroscopic properties of copper II com 9 pounds This is due to the simplicity of the delectron configuration the wide variety of stereochemistries that copper II compounds can adopt and the f xional geometric behavior that they sometimes exhibit 1 The electronic and geometric properties of a molecule are inexorably linked and this is especially true with six coordinate copper II compounds which are subject to a Jahn T ler effect However the spectral structural correlations that are sometimes d wn must often be viewed with caution as the information contained in a typical solution UV Vis absorption spectrum of a copper II compound is limited Meaningful spectral structural correlations can be obtained in a related series of compounds where detailed spectroscopic data is available In the fol 4 lowing sections two such series are examined the six coordinate CuF and 6 2 Cu H O ions doped as impurities in single

crystal hosts Using low tempera 2 6 ture polarized optical spectroscopy and electron paramagnetic resonance a very detailed picture can be drawn about the geometry of these ions in both their ground and excited electronic states We then compare the spectrosco cally determined structural data with that obtained from X ray diffraction or EXAFS measurements

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