

# Government/Types of Government

*Instructions: Answer all questions in complete sentences in the provided spaces.*

1. What is a government?

\_\_\_\_\_  
\_\_\_\_\_  
(2marks)

2. What are the four (4) types of government that exist around the world?

\_\_\_\_\_  
\_\_\_\_\_  
(4 Marks)

3. Match the phrases in Column A with the answers in Column B by writing the words in the space provided. (DO NOT DRAW LINES)

## COLUMN A

A. \_\_\_\_\_ The leader is the President, who may or may not be elected by the people.

B. \_\_\_\_\_ One man or a small clique is the leader. The leader is called a dictator, and holds power over the people by force.

C. \_\_\_\_\_ The leader inherits this position, and has a sovereign (a king, a queen, an emperor, or an empress) as leader.

D. \_\_\_\_\_ The leader is elected by the people, and is called the Prime Minister. Its people have freedom of speech and movement.

## COLUMN B

**Democracy**

**Monarchy**

**Republic**

**Dictatorship**

4. What forms of government does the Bahamas have?

\_\_\_\_\_  
(2pts)

5. Complete the chart below by writing four (4) examples of each type of government. Then turn to the blank world map on page 12 and label the countries you listed.

Types of Government	Countries Around the World
Democracy	
Dictatorship	
Monarchy	
Republic	

# Guide For Government Activity Answers

**M Walker**



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Visual Literacy in the K-12 Social Studies Classroom Matt Hensley, Stewart Waters, William B. Russell, 2023-09-01. Visual Literacy in the K-12 Social Studies Classroom is an engaging resource that unites pedagogical theory and practical strategies, empowering teachers to foster critical thinking and cultural awareness among students through the interpretation and creation of visual content. Packed with a variety of visual tools, resources, teacher-tested lesson plans and more, this book showcases the power of leveraging visual literacy to craft authentic and meaningful social studies learning experiences that resonate with learners of all ages.

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web other articles where vapour pressure curve is discussed thermodynamics the clausius clapeyron equation pressure remains equal to the vapour pressure  $p_{vap}$  as the piston moves up as long as both phases remain present all that happens is that more water turns to steam and the heat reservoir must supply the latent heat of vaporization  $\lambda$  40 65

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web mar 13 2023 the measured values  $\tau_{ign}$  is represented by the data for n decane t 1600 k at p 3 5 atm n decane t 1175 k p 5 atm n decane t 1300 k p 13 atm n decane and kerosene jet a t 1750 k p 9 atm jet a kerosene t 1320 k p 11 atm and jet a kerosene t 1380 k p 10 atm

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web a the vapor pressure curve of water intersects the p 1000 mmhg line at about 110 c this is therefore the boiling point of water at 1000 mmhg b the vertical line corresponding to 250 c intersects the vapor pressure curve of mercury at p 75 mmhg hence this is the pressure required for mercury to boil at 250 c

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web an examination of methods for calculating vapor pressure of petroleum hydrocarbon alan e zengel 1964 at high temperatures the vapor pressure of kerosene type hydrocarbons is difficult to measure accurately with laboratory apparatus be cause of difficulties in direct measurement of vapor pressure temperature

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web jan 4 2021 vapor pressure 5 mmhg at 100 f flash point 100 162 f vapor density 4 5 specific gravity 0 81 ionization potential lower explosive limit lel 0 7 upper explosive limit uel 5 nfpa health rating

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web vapour pressure of kerosene hydrocarbon the vapor pressure of a liquid is defined as the pressure exerted by the molecules that escape from the liquid to form a separate vapor phase above the liquid surface this pressure is formed in a thermodynamic equilibrium state in a closed container at a certain temperature

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web allow the gaseous phase thermodynamics to be curve fitted to 300k about 142k lower than the normal boiling point 542k 2 of kerosene nevertheless in actual calculations involving liquid kerosene fuel the heat capacity and latent heat of vaporization of liquid kerosene 12 are used for

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web as a general trend vapor pressures of liquids at ambient temperatures increase with decreasing boiling points this is illustrated in the vapor pressure chart see right that shows graphs of the vapor pressures versus temperatures for a variety of liquids

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web the vapor pressure of a liquid is defined as the pressure exerted by the molecules that escapes from the liquid to form a separate vapor phase above the liquid surface the pressure exerted by the vapor phase is called the vapor or saturation pressure vapor or saturation pressure depends on temperature

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web sep 21 2022 a vapor pressure curve is a graph of vapor pressure as a function of temperature to find the normal boiling point of liquid a horizontal line is drawn from the y axis at a pressure equal to standard pressure

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web the graph of the vapor pressure of water versus temperature in figure pageindex 3 indicates that the vapor pressure of water is 68 kpa at about 90 c thus at about 90 c the vapor pressure of water will equal the atmospheric pressure in

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web nov 1 2003 droplet evaporation modeling by the distillation curve model accounting for kerosene fuel and elevated pressures

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web vapor pressure 0 1 psi at 100 f ntp 1992 vapor density relative to air 4 5 ntp 1992 heavier than air will sink specific gravity 0 8 at 59 f uscg 1999 less dense than water will float

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web in fact when the temperature increases from 20 to 60 c the vapor pressure of kerosene rises from 4 to 18 mbar 34 this suggests that kerosene cannot be entirely vaporized at 60 c

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web kerosene icsc 0663 physical chemical information physical state appearance low viscosity liquid with characteristic odour physical dangers as a result of flow agitation etc electrostatic charges can be generated vapour pressure kpa at 37 8 c 1 3 7 exposure health effects