

CHAPTER 7 REVIEW*Chemical Formulas and Chemical Compounds***SECTION 7-1****SHORT ANSWER** Answer the following questions in the space provided.

1. C In a Stock name such as iron(III) sulfate, the roman numeral tells us _____
(a) how many atoms of Fe are in one formula unit
(b) how many sulfate ions can be attached to the iron atom
(c) the charge on each Fe ion
(d) the total positive charge of the formula unit
2. C The result of changing a subscript in a correctly written chemical formula is to _____
(a) change the number of moles represented by the formula
(b) change the charges on the other ions in the compound
(c) change the formula so that it no longer represents the compound it previously represented
(d) have no effect on the formula
3. The explosive TNT has the molecular formula $C_7H_5(NO_2)_3$.
4 elements a. How many elements make up this compound?
6 oxygen atoms b. How many oxygen atoms are present in one molecule of $C_7H_5(NO_2)_3$?
21 atoms c. How many atoms in total are present in one molecule of $C_7H_5(NO_2)_3$?
 4.2×10^{24} atoms d. How many atoms are present in a sample of 2×10^{23} molecules of $C_7H_5(NO_2)_3$?
4. How many atoms are present in each of these formula units?
11 atoms a. $Ca(HCO_3)_2$
45 atoms b. $C_{12}H_{22}O_{11}$
10 atoms c. $Fe(ClO_2)_3$
9 atoms d. $Fe(ClO_3)_3$
5. N_2O_5 a. What is the formula for the compound dinitrogen pentoxide?
carbon(IV) sulfide b. What is the Stock name for the covalent compound CS_2 ?
 H_2SO_3 c. What is the formula for sulfurous acid?
phosphoric acid d. What is the name for the acid H_3PO_4 ?

Guided And Study Chemical Names And Formulas

Clifford C. Houk, Richard Post



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CHEMISTRY HANDBOOK & STUDY GUIDE Gr11-12 NE Kevin Smith,2024-02-01 A comprehensive summary of Grade 11 12 Physics Simple logical summaries with example exam questions and work through solutions The book covers the fundamentals of Grade 11 12 Physics and complements the material in any class text *Chemistry, Student Study Guide* James E. Brady,Fred Senese,2008-01-28 The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon as it forms around particles of a metal catalyst Carbon nanotubes are a recently discovered allotrope of carbon Three other allotropes of carbon buckyballs graphite and diamond are illustrated at the left as is the molecule methane CH_4 from which nanotubes and buckyballs can be made The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement **A**

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Industrial and Engineering Chemistry , 1925 **Industrial & Engineering Chemistry** , 1923

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