

1. FeSO_4 _____

2. $\text{Fe}_2(\text{SO}_4)_3$ _____

3. Fe_2O_3 _____

4. Fe_3O_4 _____

ANSWERS

NAME: _____ & FORMULARS: _____

ANSWERS

a. Write the formulas for the following ionic compounds:

- (1) sodium chloride _____
- (2) potassium nitrate _____
- (3) calcium chloride _____
- (4) hydrogen iodide _____
- (5) sodium carbonate _____
- (6) hydrogen fluoride _____
- (7) ammonia _____
- (8) ammonium chloride _____

b. Write the names for the following ionic compounds:

- (1) FeCl_2 _____
- (2) Fe_2O_3 _____
- (3) FeSO_4 _____
- (4) $\text{Fe}_2(\text{SO}_4)_3$ _____
- (5) Fe_3O_4 _____
- (6) Fe_2O_3 _____
- (7) Fe_2O_3 _____
- (8) Fe_2O_3 _____
- (9) Fe_2O_3 _____

c. Write the ionic names of the following ionic compounds:

- (1) FeCl_2 _____
- (2) Fe_2O_3 _____
- (3) FeSO_4 _____
- (4) $\text{Fe}_2(\text{SO}_4)_3$ _____
- (5) Fe_3O_4 _____

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 - Chapter 2: Essential Elements of Holt Chemistry Covalent Compounds Review Answers
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 - Chapter 4: Holt Chemistry Covalent Compounds Review Answers in Specific Contexts
 - Chapter 5: Conclusion
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