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کامیاب ہو کر

5.5.C - (Part-B-A- 2013)  
(New Course, Old Scheme)  
Session: 2011-2013

Paper Code

1

4

8

6

## Chemistry (Objective Type)

## گروپ-II

## کیمیشتری (مبروضی)

Marks: 12

وقت: 15 منٹ Time: 15 Minutes

نمبر: 12

نوٹ: تمام سوالات کے جوابات دی گئی سرروضی جوابی کاپی پر لکھئے۔ ہر سوال کے چار ممکنہ جوابات A, B, C, D دیے گئے ہیں۔ جس میں جواب کو آپ درست سمجھیں، جوابی کاپی پر اس سوال نمبر کے سامنے جزو A, B, C, D کے دائروں میں سے صحیح دائرے کو مار کر یا پین کی مدد سے لکھ دیں۔

NOTE: Write answers to the questions on objective answer sheet provided. Four possible answers A, B, C & D to each question are given. Which answer you consider correct, fill the corresponding circle A, B, C or D given in front of each question with Marker or pen ink on the answer sheet provided.

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1.1. The volume of water on freezing becomes:

(A) increase زیادہ

(B) less کم

(C) normal نارمل

1.1. پانی کو فریز کرنے سے اس کا حجم بڑھتا ہے۔

(D) dense گہرا

2. The quantity of copper in coinage silver is:

(A) 20%

(B) 10%

(C) 100%

2. نکلے جانے والی چاندی میں کاپر کی مقدار ہوتی ہے۔

(D) 1000%

3. Oxidation number of 'Mn' in  $\text{KMnO}_4$  is:

(A) 6

(B) 5

(C) 7

3.  $\text{KMnO}_4$  میں 'Mn' کا آکسیدیشن نمبر ہے۔

(D) 3

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4. Sodium hydroxide is a powerful:

(A) alkali آگلی

(B) acid تیزاب

(C) metal دھات

4. سولیم ہائیڈروآکسائیڈ ایک طاقتور ہے۔

(D) salt نمک

5. The melting point of mercury is:

(A)  $50^\circ\text{C}$ (B)  $-38.83^\circ\text{C}$ (C)  $25^\circ\text{C}$ 

5. مرکری کا پگھلاؤ پوائنٹ ہے۔

(D)  $0^\circ\text{C}$ 6. The molecular mass of sulphuric acid ( $\text{H}_2\text{SO}_4$ ) is:

(A) 98 gm

(B) 98 u

(C) 9.8 gm

6. سلفورک ایسڈ ( $\text{H}_2\text{SO}_4$ ) کا مالیکیولر ماس ہے۔

(D) 9.8 a.m.u

7. A metalloid is:

(A) Bi

(B)  $\text{H}_2\text{O}$ 

(C) Zn

7. دھات لٹائے۔

(D)  $\text{CO}_2$ 

8. The atomic number of hydrogen is:

(A) 2

(B) 1

(C) 4

8. ہائیڈروجن کا ایٹمی نمبر ہے۔

(D) 3

9. X-Rays were discovered in:

(A) 1910 A.D.

(B) 1911 A.D.

(C) 1912 A.D.

9. ایکس ریز (X-Rays) دریافت ہوئیں۔

(D) 1913 A.D.

10. Number of groups in periodic table are:

(A) 7

(B) 9

(C) 18

10. دوری جدول میں گروپس کی تعداد ہے۔

(D) 17

11. The bond present in sodium and chlorine is:

(A) Single covalent bond

(B) Covalent bond

11. سوڈیم اور کلورین میں موجود بانڈ ہے۔

(C) ionic bond

(D) Metallic bond

12. The outer shell of carbon atom has electrons:

(A) 1

(B) 2

(C) 3

12. کاربن ایٹم کے بیرونی شیل میں الیکٹرونز ہوتے ہیں۔

(D) 4

# Guide Of Chemistry 2015 Class 9th

**Pam Hazelton, Brian Murphy**



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