



ANSWER GUIDE
PHYSICAL SCIENCES: GRADE 12
CLASS TEST: ACIDS & BASES

Written on: 25.06.2020

Total Marks: 65

1	1.1	B ✓✓	2
	1.2	C ✓✓	2
2	Hydrochloric acid is a strong acid and sodium hydroxide is a strong base, ✓ therefore the equivalence point will be reached at a pH close to 7, ✓ which is in the colour change range of bromothymol blue.		2
3	3.1	An acid is a proton donor. ✓	1
	3.2	A concentrated acid is an acid solution with a high proportion of acid (solute) to water (solvent). ✓✓ regardless of the degree of ionisation or dissociation of the acid in the solution. A strong acid is an acid that dissociates or ionises almost completely when in solution, regardless of its concentration. ✓✓	4
4	4.1	LiOH ✓	5
	4.2	H ₂ PO ₄ ✓	
	4.3	HNO ₃ ✓	
	4.4	HCOOH ✓	
	4.5	NaHCO ₃ ✓	
5	$\text{H}_2\text{SO}_4 + 2\text{H}_2\text{O} \checkmark \rightarrow 2\text{H}_3\text{O}^+ \checkmark + \text{SO}_4^{2-} \checkmark$ OR: $\text{H}_2\text{SO}_4 + \text{H}_2\text{O} \rightarrow \text{H}_3\text{O}^+ + \text{HSO}_4^-$		3
6	6.1	A standard solution is a solution of known concentration. ✓✓	2
	6.2	$c = \frac{m}{MV} \quad \therefore m = cMV \checkmark$ $= (0,15)(106)(0,25) \checkmark$ $= \underline{3,975 \text{ g Na}_2\text{CO}_3} \checkmark$	3
	6.3	$\frac{n_a}{n_b} = \frac{c_a V_a}{c_b V_b} \quad \therefore c_a = \frac{n_a c_b V_b}{n_b V_a} \checkmark$ $= \frac{(1)(0,15)(25)}{(1)(27,5)} \checkmark$ $= \underline{1,38 \times 10^{-1} \text{ mol.dm}^{-3}} \checkmark$	4
7	pH = $-\log[\text{H}^+] \checkmark$ and: HCl dissociates completely $\therefore 0,2 \text{ mol.dm}^{-3} \text{ H}^+$ ions = $-\log[0,2] \checkmark$ = $0,69 \checkmark$		3

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Huangqi Zhang



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