

Chemistry (B) HW Chapter 13 (April 1 Due April 16 Test April 17)

Matching

Match each item with the correct statement below.

- | | |
|------------------|-------------------------|
| a. melting point | d. evaporation |
| b. boiling point | e. vaporization |
| c. sublimation | f. normal boiling point |

- _____ 1. vaporization at the surface of a liquid that is not boiling
- _____ 2. the conversion of a liquid to a gas below the boiling point
- _____ 3. the temperature at which the vapor pressure of a liquid is equal to the external pressure
- _____ 4. the temperature at which the vapor pressure of a liquid is equal to 1 atmosphere
- _____ 5. the temperature at which a solid changes into a liquid
- _____ 6. the change of a solid directly to a vapor

Match each item with the correct statement below.

- | | |
|-------------------------|-------------------|
| a. kinetic theory | d. barometer |
| b. atmospheric pressure | e. kinetic energy |
| c. vapor pressure | |

- _____ 7. All matter consists of tiny particles that are in constant motion.
- _____ 8. the energy an object has due to its motion
- _____ 9. the pressure resulting from the collision of atoms and molecules with objects
- _____ 10. a measure of the force exerted by a gas above a liquid

Multiple Choice

Identify the choice that best completes the statement or answers the question.

- _____ 11. According to the kinetic theory, collisions between molecules in a gas _____.
 - a. are perfectly elastic
 - b. are inelastic
 - c. never occur
 - d. cause a loss of total kinetic energy
- _____ 12. Which of the following statements is part of the kinetic theory?
 - a. The particles of a gas move independently of each other.
 - b. The particles in a gas move rapidly.
 - c. The particles in a gas are relatively far apart.
 - d. all of the above
- _____ 13. The average speed of oxygen molecules in air is about _____.
 - a. 0 km/h
 - b. 170 km/h
 - c. 1700 km/h
 - d. 17,000 km/h
- _____ 14. Which of the following statements is NOT true, according to the kinetic theory?
 - a. There is no attraction between particles of a gas.
 - b. Only particles of matter in the gaseous state are in constant motion.
 - c. The particles of a gas collide with each other and with other objects.
 - d. All of the statements are true.

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M Mosston



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