Realidades 📵 🛢

Capítulo 3B

Practice Workbook Answers

349-1

Answers will vary:

La ensalada de frutas: list of fruits La carne: list of meats. Las verduras: list of vegetables Bebemos: list of beverages

Answers will vary.

348-2

- 1. la carne
- 2. las viendiunas
- 3. Bas firested
- 4. has oprasas.

Answers may vary.

- 1. Debemos comer las uvas para mantener la salud.
- 2. La ensalada de frutas conpilátanos es sabrosa.
- Comemos la manteguilla con el pan tostado.
- 4. Bebemos el agua para mantener the published.

Answers will vary.

S48-30

Answers will vary. Possible answers

- 1. Si, el tomate es bueno para la stallusch.
- 2. Camino todos los días para mantener la salud.
- 3. No. la mantequilla no es buena para la saludi.
- 4. Estor de acuerdo
- 5. Debo caminar todos los días para mantener la salud.
- Prefiero levantar pesas.
- No estoy de acuerdo.

- —los quisantes.
 - no me gustan los guisantes
- 2. -el arroz / pollo
 - -- Prefiero el arroz con pollo

- los espaguetis.
 - -me gustan los espaguetis
- el belado
 - -- Prefiero el helado
- —el pescado: -como el pescado
- partieles
 - no como pasteles en el almiuerzo.
- bistec / papas
 - -me gusta el bistec con papas

349-5

A

Row 1: _____, sabrosos, sabrosa, salbrosas

Row 2: práctico, _____, práctica, prefetticas.

Row 3: fácil, fáciles, ____ Row 4: aburrido, ____, aburrida,

aburridas. Row 5: difficit, difficites, difficit, ... Row 6: ---- , divertidos,

divertida, divertidas Row 7: artístico, artísticos,

artisticas Row 8: bueno, buenos,

_, trabajadores. trabajadora, trabajadoras

Answers may vary. Any adjective given should agree in gender and number with answers given below:

- 1. bearens
- 2. Michles
- 3. difficili
- 4. prácticas
- 5. sabroso
- 6. artistions
- trabajadores.
- B. divertidas

348-6

- 1. es / Es artístico-
- 2. es / fis perezosa
- 3. son / Son estudiosas
- 4. son / Son sociables
- S. somos / Somos (son) trabaladoras
- 6. soy / Soy (eres) deportista

388-7

- 1. Para mantener la salud, Eva come muchas verduras y frutas cada
- 2. Si, ella hace ejercicio.
- 3. Si, a ella le gustan las frutas.
- 4. Eva prefiere levantar pesas para mantener la salud.
- El agua es la bebida favorita de From:
- No debernos comer pasteles. porque son malos para la stabled).

Crucigrama (38-8)

Acress:

- 3. elercicio
- 5. pescado-
- lechuga. 8. polici
- 10, salud
- 1.2. behirden
- 13. across
- 16. algo-
- 18. hambee
- 20, acuerdo
- 22. pasteles
- 24. debes

Distriction:

- 1. carne
- 2. transaction
- 4. jurdilars
- 7. espaguetis
- 9. mantener
- 11. LOCKER
- 14. zamahorias
- 15, subrosa.
- 17. heliado.
- 19. ceholia
- 21. cena
- 23. todos

Organizer (38-9)

I. Vocabulary Answers will vary. III. Grammar

- 1. singular / plural
- 2. -s. -es
- 3. col. 1. col. 2. NOW. somos.
 - COMMON. 616

signis. sion.

Guided 26 Answers

Timothy Rasinski, Nancy Padak, Rick
M. Newton, and Evangeline Newton

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web in today s experiment you will determine the solubility product ksp of calcium hydroxide ca oh 2 by measuring the concentration of ca oh 2 in a saturated solution calcium hydroxide is a sparingly soluble salt that dissolves according to the following reaction the solubility product expression for this reaction is

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web expert verified chem 182 determining the ksp of calcium hydroxide calcium hydroxide is an ionic solid that is sparingly

soluble in water a saturated aqueous solution of ca oh is represented in equation form as shown below ca oh s ca aq 2oh aq the solubility product expression describes in mathematical terms the equilibrium

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web ksp for calcium hydroxide calcium hydroxide is a soft white caustic powder used in making mortar cements calcium salts paints and petrochemicals it is also used in saltwater aquaria to make up kalkwasser limewater solutions for reef tanks and is used as a ph regulating agent notice that calcium hydroxide is divalent and thus has twice

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web use the concentration of calcium hydroxide to calculate the concentration of calcium ions and hydroxide ions in the solution ca oh 2 à ca 2 2oh 5 x 10 2 m calcium ions 5 x 10 2 m 2 0 m hydroxide ions calculate calcium hydroxide ksp using the concentrations of calcium ions and hydroxide ions

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web madison bowe elana keller april 15 2014 introduction the objective of this lab was to measure the ksp of a slightly soluble basic salt and how solubility changes with temperature ksp is another form of equilibrium expression that deals with solubility the reaction that will take place is ca oh 2 ca2 2 oh

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web the chemical equation used was ca oh 2 s ca2 aq 2oh aq the solubility product was determined to be ksp ca2 oh from the titration which was made by dripping a concentrated solution of calcium hydroxide into hydrochloric acid the calcium hydroxide was filtered before starting the titration

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web jul 12 2023 because the concentration of a pure solid such as ca 3 po 4 2 is a constant it does not appear explicitly in the equilibrium constant expression the equilibrium constant expression for the dissolution of calcium phosphate is therefore k ca2 3 po 3 4 2 ca3 po 4 2 ca3 po 4 2 k ksp ca2 3 po 3 4 2

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web lab report 8 determinin of calcium hydroxide the purpose of this was to titrat e a saturated ca oh 2 soultion with a st andard hcl solution and find the ksp or the oh concentration was f ound and then the concentration was found and u for this experiment a titration of calcium hydroxide with hydrochlo

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web question determining the ksp of calciumhydroxide ca oh 2 s ca2 aq 20h aq for this experiment the titration of calciumhydroxide with hydrochloric acid wasinvestigated determining the ksp of calcium hydroxide ca oh 2 s ca2 aq 20h aq for this experiment the titration of calcium

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web question chem 182 experiment 8 determining the ksp of calcium hydroxide calcium hydroxide is an ionic solid that is sparingly soluble in water solution of ca oh 2 is represented in equation form lution of ca on s nronic solid that is sparingly soluble in water a saturated aqueous as shown below

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web therefore it was determined through error calculations and ease of usethat the most reliable valid method to use to get the closest ksp value tothe theoretical ksp for ca was the determination through ph method 2 introduction calcium hydroxide is a strong base soluble in water

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web plasma membrane homework symbiosis practice worksheet benzoin condensation grade a determining the ksp of calcium hydroxide after reading the lab complete items a b c and d title purpose chemicals and equipment and summary of procedure as described on page 10 of exp 1 on an 8 1 2 x 11 sheet of paper

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web we will filter the solid calcium hydroxide from the solution and then titrate the dissolved calcium hydroxide in solution with a known concentration of hydrochloric acid using titrimetric calculations we will determine the ca 2 and oh concentrations in solution

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web calculate the ksp for calcium hydroxide ksp ca2 oh 2 0 0 2 ksp 5 x 10 4 find the accepted value of the ksp for calcium hydroxide and compare it with your value discuss the discrepancy and suggest possible sources of experimental error accepted value for ksp for sodium hydroxide is $5 \times 10 \times 10 \times 10^{-5}$

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