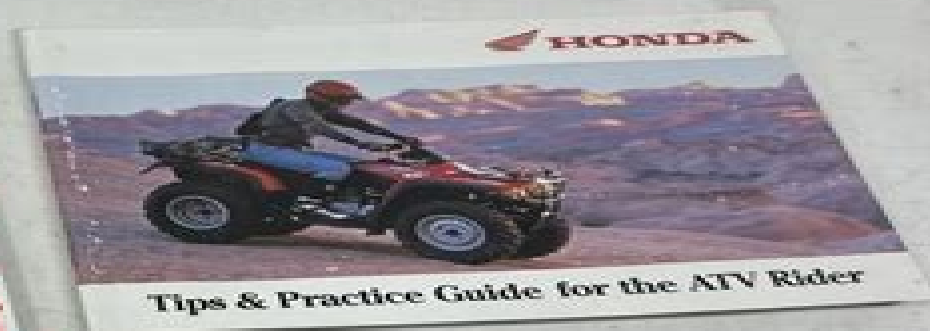




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web solution average atomic mass of Li 6 941 amu $\times$ 6 0151 amu $\times$ 7 0160 amu 6 941 1 0009 $\times$ 7 0160 1 0009 $\times$ 0 075 $\times$ 0 075 $\times$ 0 075 corresponds to a natural abundance of 6Li of 7 5 percent the natural abundance of 7Li is $\times$ 0 925 or 92 5 percent 3 7 the unit factor required is 6 022

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web a molar mass Li_2CO_3 2 6 941 g 12 01 g 3 16 00 g 73 89 g b molar mass CS_2 12 01 g 2 32 07 g 76 15 g c molar mass $CHCl_3$ 12 01 g 1 008 g 3 35 45 g 119 37 g d molar mass $C_6H_8O_6$ 6 12 01 g 8 1 008 g 6 16 00 g 176 12 g e molar mass KNO_3 39 10 g 14 01 g 3 16 00 g 101 11 g

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