The 12 Principle of



Green Chemistry With A Corresponding Green Nanotechnology Approach



Green chemistry is a method of implementing green nanotechnology that attempts to increase efficiency while minimizing negative consequences for environmental contamination. While no reaction can be completely "green," the overall negative impact of nanotechnology research and industry can be mitigated by following the 12 Principles of Green Chemistry wherever applicable.

1. Prevent waste.





Develop novel 'greener' chemical nanomaterial syntheses that decrease or eliminate waste streams and products at the source by including pollution avoidance into the design.



Produce nanomaterial with the necessary physicochemical characteristics and less or no toxicity

3. Design less hazardous chemical.

4. Use renewable feedstocks



Work towards developing syntheses that use environmentally friendly reagents and solvents to produce compounds with little or no toxicity to humans and the environment.



sustainable and environmentally friendly raw materials and feedstock.

5. Utilize catalysts

Avoid chemical derivatives



Utilize nanocatalysts produced in a green manner to reduce waste. Develop catalytic techniques that can improve reaction selectivity and overall process efficiency significantly.



If at all possible, avoid employing blocking or protective groups, as well as any temporary alterations. Utilize selective nanosyntheses to eliminate solvent-intensive syntheses.

Maximize atom economy

8. Use safer solvents



Design and develop syntheses such that the end product includes as much of the raw ingredients as possible.



Hazardous solvents, separating agents, and other auxiliary chemicals should be avoided. To improve materials efficiency and minimize processes, use a bottom-up strategy.

Increase energy efficiency

Design chemicals and products



Develop synthesis methods that, if possible, are carried out at room temperature and pressure. Develop synthesis methods that, if possible, are carried out at room temperature and



Assess and quantify the manomaterial's behavior in the environment, study their fate and major chemical interactions with other particles or organisms. Design nanomaterial's to degrade

11. process control to prevent

Minimize the accidents



Develop a nanoscience Life Cycle Assessment (LCA), which is a critical tool for controlling risks in nanoscience research, development, and commercialization.



Minimize the potential for chemical accidents including explosions, fires, and releases into the environment.

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Chemistry of Pyrotechnics John A. Conkling, Chris Mocella, 1985-07-02 A perennial bestseller Chemistry of Pyrotechnics and Explosives Basic Principles and Theory is simply the most definitive reference in this field Author J A Conkling first covers the requisite background in chemistry thermodynamics and light emission introduces oxidizing agents fuels binders and retardants then explores virtually every aspect of formulating pyrotechnics Topics include the requirements for and preparation of high energy mixtures ignition and propagation heat and delay compositions and color and light production including sparks flitter and glitter The journal Pyrotechnica said this book belongs on every pyrotechnist s bookshelf **Principles of Pyrotechnics** A. A. Shidlovskiv, 1964 Pyrotechnics Fraunhofer-Institut für Treib- und Pyrotechnics International Pyrotechnics Society. Steering Committee, Explosivstoffe, 1985 **Pyrotechnics**, Basic **Principles, Technology, Application**, 1995 Engineering Design Handbook United States. Army Materiel Handbook of Pyrotechnics Karl O. Brauer, 1974-07-07 This handbook provides extremely useful data Command.1963 and information about theory and industrial applications that formerly were limited to applications in military ordnance fireworks and rock blasting These civilian applications include spacecraft aircraft underwater vehicle systems metal forming cladding riveting and some other production methods Also pairs well with Fireworks Principles and Practice 4th Edition ISBN 0820604070 Military and Civilian Pyrotechnics ISBN 0820603643 Pyrotechnics ISBN 0820601128 Visit us at www chemical **Chemistry of Pyrotechnics** John A. Conkling, 2010 Chemistry of Pyrotechnics Second Edition is simply publishing com the most definitive reference in this field Author J A Conkling covers the requisite background in chemistry thermodynamics and light emission introduces oxidizing agents fuels binders and retardants and then explores virtually every aspect of formulating pyrotechnics Topics include the requirements for and preparation of high energy mixtures ignition and propagation heat and delay compositions and color and light production including sparks flitter and glitter Pyrotechnics Fred Volk, 1995 Military and Civilian Pyrotechnics Herbert Ellern, 1968-05-30 Written for manufacturers laboratory pyrotechnicians and lay students this book contains an enormous amount of invaluable data on the properties and production of pyrotechnic materials on heat and light production on aerosols energy and noise Includes a manufacturing formulary and a good reference section Its wealth of information and highly readable style make it a must have in every pyrotechnician s library These titles may also pair well with these books Brauer Karl O Handbook of Pyrotechnics ISBN 978 0 8206 0349 0 Rev Ronald Lancaster Fireworks Principles and Practice 4th Edition ISBN 0 8206 0407 0 Weingart George W Pyrotechnics ISBN 0 8206 0112 8 Visit us at www chemical publishing com Fireworks Ronald Lancaster, 1972 Contents 1 The History

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