

Isotopes Practice Set

- What does atomic number tell you about an element?
- What does the number used to isotopes signify?
- How can you tell isotopes apart?

For each of the following isotopes, write the number of protons, neutrons, and electrons.

	Chlorine-35	Chlorine-37
# of protons	17	17
# of neutrons	18	20
# of electrons	17	17

	Nitrogen-14	Nitrogen-15
# of protons	7	7
# of neutrons	7	8
# of electrons	7	7

	Carbon-12	Carbon-13
# of protons	6	6
# of neutrons	6	7
# of electrons	6	6

	Oxygen-16	Oxygen-18
# of protons	8	8
# of electrons	8	8
# of neutrons	8	10

Fill in the blank isotopes names including chemical symbol, atomic number, and mass number from the chart. Use your periodic table and the information provided.

	Isotope: $^{12}_6\text{C}$	Isotope: $^{16}_8\text{O}$
# of protons	6	8
# of neutrons	6	8
# of electrons	6	8

	Isotope: $^{14}_7\text{N}$	Isotope: $^{23}_{11}\text{Na}$
# of protons	7	11
# of neutrons	7	12
# of electrons	7	11

	Isotope: $^{14}_7\text{N}$	Isotope: $^{16}_8\text{O}$
# of protons	7	8
# of neutrons	7	8
# of electrons	7	8

	Isotope: $^{12}_6\text{C}$	Isotope: $^{13}_6\text{C}$
# of protons	6	6
# of neutrons	6	7
# of electrons	6	6

Isotope Practice Set Answers

Disha Experts

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the 'Disha Experts' text.

Isotope Practice Set Answers:

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