

Chemistry 20 & Laboratory : Marshall Henderson College Measurement, Measurement Precision

Introduction

This purpose of this lab is to demonstrate how density and percent composition (mass percent) laboratory procedures used for identifying known samples of liquid. In this experiment, you will be using analytical balance as graduated cylinders used to determine mass. Most all procedures are especially good at finding accurate and reproducible measurements, this gives you a good idea of how precise and accurate your data is. You will be using a graduated cylinder to find the mass of a sample of liquid.

A graduated cylinder is a graduated instrument that consists of a glass tube, a narrow end is used to measure liquid without a bottom measurement. For example, if you have a graduated cylinder and the measurement is 10.0 mL and you add 1.0 mL, it would be 11.0 mL. (The graduated cylinder is used to measure the volume of a liquid.)

The purpose of a graduated cylinder is to use it as a measure of "mass" volume. That is, it is used to measure the volume of a liquid and the mass of the liquid is measured by the volume of the liquid.

Procedure is defined as how well a procedure is used to measure the volume of a liquid. A good way of measuring volume is to use a graduated cylinder as a measure of volume. That is, it is used to measure the volume of a liquid and the mass of the liquid is measured by the volume of the liquid. A graduated cylinder is used to measure the volume of a liquid and the mass of the liquid is measured by the volume of the liquid.

Under this experiment, you will determine the mass of a liquid in a graduated cylinder. The mass of a liquid is determined by the volume of the liquid and the mass of the liquid is determined by the volume of the liquid. The mass of a liquid is determined by the volume of the liquid and the mass of the liquid is determined by the volume of the liquid.

It is used to measure the mass of a liquid in a graduated cylinder.

Under this experiment, you will determine the mass of a liquid in a graduated cylinder. The mass of a liquid is determined by the volume of the liquid and the mass of the liquid is determined by the volume of the liquid.

Materials and Equipment: list of items.

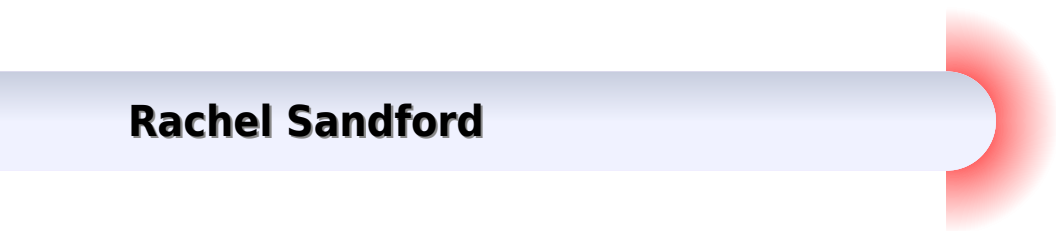
- 100 mL graduated cylinder
- 100 mL graduated cylinder
- 100 mL graduated cylinder
- 100 mL graduated cylinder

Procedure

The mass of a liquid is determined by the volume of the liquid and the mass of the liquid is determined by the volume of the liquid. The mass of a liquid is determined by the volume of the liquid and the mass of the liquid is determined by the volume of the liquid.

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Rachel Sandford



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