

Experiment 15: Specific Heat of a Metal

Purpose: To determine the specific heat of a substance.

Procedure:

Record all data in Data Table 1.

1. Heat 250 mL of water in a 400-mL beaker until it is boiling gently.
2. While the water is heating, determine and record the mass of a clean, dry 50-mL beaker to the nearest 0.01 g. Add between 80 g and 120 g of lead shot to the beaker. Measure the combined mass of the beaker and lead to the nearest 0.01 g and record the measurement.
3. Transfer the lead shot to a large, dry test tube. Use the utility clamp to suspend the test tube in the boiling water as shown in Figure 15.1. The lead shot should be below the level of the water in the beaker. Leave the test tube in the boiling water bath for at least 10 minutes.
4. While the lead shot is heating, measure 100 mL of distilled water in a graduated cylinder. Pour the water into a Styrofoam cup and place the cup in a 250-mL beaker for support.
5. Measure and record the temperature of the water in the Styrofoam cup and of the water in the boiling bath.
6. Remove the test tube from the boiling water and quickly pour the lead shot into the water-filled, plastic-foam cup. Place a thermometer and a glass stirring rod into the cup. Use the stirring rod to gently stir the lead shot. *Do not stir the shot with the thermometer.* Note the temperature frequently and record the maximum temperature reached.
7. Pour the water off and return the lead shot to your teacher.
8. Follow the same procedure to determine the specific heat of a second metal.

Data Table 1:

Measurements of Mass and Temperature		
	Trial 1 (Pb)	Trial 2 (Al)
Mass of metal shot (g)	55.7	68.1
Mass of water in cup (g)	100.0	100.0
Initial temperature of water in cup ($^{\circ}\text{C}$)	22.0	23.0
Initial temperature of metal shot ($^{\circ}\text{C}$) (same as temperature of boiling water)	101.1	99.8
Maximum temperature of metal + water	27.1	32.0

Analysis Questions:

1. Determine the changes in temperature of the water (ΔT_{water}) and of the lead shot (ΔT_{metal}) for each trial.
2. Calculate the heat gained by the water in each trial.

Lab 22 Specific Heat Of Metals Answers

Ying Liu



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Table of Contents Lab 22 Specific Heat Of Metals Answers

1. Understanding the eBook Lab 22 Specific Heat Of Metals Answers
 - The Rise of Digital Reading Lab 22 Specific Heat Of Metals Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Lab 22 Specific Heat Of Metals Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Lab 22 Specific Heat Of Metals Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Lab 22 Specific Heat Of Metals Answers
 - Personalized Recommendations
 - Lab 22 Specific Heat Of Metals Answers User Reviews and Ratings
 - Lab 22 Specific Heat Of Metals Answers and Bestseller Lists
5. Accessing Lab 22 Specific Heat Of Metals Answers Free and Paid eBooks
 - Lab 22 Specific Heat Of Metals Answers Public Domain eBooks
 - Lab 22 Specific Heat Of Metals Answers eBook Subscription Services
 - Lab 22 Specific Heat Of Metals Answers Budget-Friendly Options
6. Navigating Lab 22 Specific Heat Of Metals Answers eBook Formats

- ePub, PDF, MOBI, and More
 - Lab 22 Specific Heat Of Metals Answers Compatibility with Devices
 - Lab 22 Specific Heat Of Metals Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Lab 22 Specific Heat Of Metals Answers
 - Highlighting and Note-Taking Lab 22 Specific Heat Of Metals Answers
 - Interactive Elements Lab 22 Specific Heat Of Metals Answers
 8. Staying Engaged with Lab 22 Specific Heat Of Metals Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Lab 22 Specific Heat Of Metals Answers
 9. Balancing eBooks and Physical Books Lab 22 Specific Heat Of Metals Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Lab 22 Specific Heat Of Metals Answers
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Lab 22 Specific Heat Of Metals Answers
 - Setting Reading Goals Lab 22 Specific Heat Of Metals Answers
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Lab 22 Specific Heat Of Metals Answers
 - Fact-Checking eBook Content of Lab 22 Specific Heat Of Metals Answers
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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