

CORE LABORATORY MANUAL OF CHEMISTRY

For Class XI



GOYAL BROTHERS PRAKASHAN

Goyal Brothers Prakashan Chemistry Lab Manual Class

**Neena Sinha, R Rangarajan, R P
Manchanda, R K Gupta, Rajesh Kumar**

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Core Science Lab Manual with Practical Skills for Class IX V. K. Sally, Chhaya Srivastava, Goyal Brothers Prakashan, 2019-01-01 Goyal Brothers Prakashan **Core Science Lab Manual with Practical Skills for Class X** V. K.

Sally, Chhaya Srivastava, Goyal Brothers Prakashan, 2019-01-17 Goyal Brothers Prakashan **The Indian National**

Bibliography, 2018-04 **A New Approach to ICSE Chemistry for Class IX (A.Y. 2023-24) Onward** V. K.

Sally, 2023-05-20 The book *A New Approach to ICSE Chemistry for Class IX* has been written in accordance with the latest syllabus prescribed by the Council for Indian School Certificate Examination New Delhi. The book is divided into five distinct parts: 1 The first part constitutes chapters on The Languages of chemistry, Chemical Changes and Reactions, and Water to give an idea to the students about the diversity of matter; 2 The second part constitutes chapters on Atomic Structure and Chemical Bonding and The Periodic Table. This part will help the students to understand unity in diversity of matter. It further explains the rules required for reaction between various forms of matter; 3 The third part constitutes chapter on Study of the First Element Hydrogen and Study of Gas Laws. This will familiarise the students with element hydrogen and their chemical properties. It further explains the behaviour of gases under changes of temperature and pressure and also provides explanation in terms of molecular motion; 4 The fourth part deals with Atmospheric Pollution; 5 The fifth part deals with Practical Chemistry. The salient features of the book: Large number of chemical reactions are described with experimental observations. Important points have been highlighted for location of precise answers. All definitions and other important points are highlighted. Diagrams are made simple and more informative. In the chapter on the language of chemistry, a new method has been introduced for balancing chemical equations. This method is based on atomic numbers, which is a fundamental property of the elements. A chapter on Practical Chemistry is added to help students for laboratory work. Questions have been added chapter wise under the heading Question Bank for the benefit of students. The solution can be accessed through QR Code given at the end of each chapter. Periodic Table showing mass number, atomic number of various elements along with list of actual names of the elements is provided. A Specimen Paper Solved has also been included through QR Code for the benefit of students. Annual Examination Paper 2019 Solved has been given through QR Code. Model Test Papers 1 to 5 Unsolved have been given through QR Code. We hope this book will prove very useful to the students and teachers. Suggestions and constructive criticism for the further improvement of the book would be gratefully acknowledged.

Author **Practical/Laboratory Manual Chemistry Class - XI** Er. Meera Goyal, 2021-05-29 1 Basic Laboratory

Techniques: 1 To cut a glass tube or glass rod; 2 To bend the glass rod at an angle; 3 To draw a glass jet from a glass tube; 4 To bore a cork and fit a glass tube into it. Viva Voce: 2 Characterisation and Purification of Chemical Substances: 1 To determine the melting point of the given unknown organic compound and its identification; simple laboratory technique; Viva Voce: 2 To determine the boiling point of a given liquid when available in small quantity; simple laboratory method; Viva Voce: 3 To

prepare crystals of pure potash alum $K_2SO_4 \cdot Al_2(SO_4)_3 \cdot 24H_2O$ from the given impure sample 4 To prepare the pure crystals of copper sulphate from the given crude sample 5 To prepare pure crystals of benzoic acid from a given impure sample Viva Voce 3 Measurement of pH Values 1 To determine the pH value of vegetable juices fruit juices tap water and washing soda by using universal pH paper 2 To determine and compare the pH values of solutions of strong acid HCl and weak acid CH_3COOH of same concentration 3 To study the pH change in the titration of strong base Vs strong acid by using universal indicator paper 4 To study the pH change by common ion CH_3COO^- ion in case of weak acid CH_3COOH 5 To determine the change in pH value of weak base NH_4OH in presence of a common ion NH_4^+ Viva Voce 4 Chemical Equilibrium 1 To study the shift in equilibrium between ferric ions and thiocyanate ions by changing the concentrations of either of the ions 2 To study the shift in equilibrium between Co^{2+} and Cl^- ions by changing the concentrations of either of the ions Viva Voce 5 Quantitative Analysis 1 To prepare M 10 oxalic acid solution by direct weighing method 2 To prepare M 10 solution of sodium carbonate by direct weighing method 3 To determine the strength of given solution of sodium hydroxide by titrating it against N 10 or M 20 solution of oxalic acid 4 To determine the strength of a given solution of hydrochloric acid by titrating it against a standard N 10 or M 20 sodium carbonate solution Viva Voce 6 Qualitative Analysis Analysis of Anions Analysis of Cations Viva Voce 7 Detection of Elements in Organic Compounds 1 To detect the presence of nitrogen sulphur and halogens in a given organic compound by Lassaigne's test 2 To detect the presence of nitrogen sulphur and halogens in the given organic compound sample number by Lassaigne's test Viva Voce INVESTIGATORY PROJECTS 1 Checking of Bacterial Contamination in Water 1 To check the bacterial contamination in drinking water by testing sulphide ions Viva Voce 2 Methods of Water Purification 1 To purify water from suspended impurities by using sedimentation 2 To purify water by boiling 3 To purify water by distillation method 4 To purify water by reverse osmosis technique 5 To purify water by GAC method 6 To purify water by bleach treatment 7 To purify water by oxidising agent 8 To purify water by ozone treatment method Viva Voce 3 Water Analysis 1 To test the hardness of different water samples Viva Voce 4 Foaming Capacity of Various Soaps 1 To compare the foaming capacity of different washing soaps 2 To study the effect of addition of sodium carbonate on foaming capacity of washing soap Viva Voce 5 Tea Analysis 1 To study the acidity of different samples of tea leaves tea by using pH paper Viva Voce 6 Analysis of Fruits and Vegetable Juices 1 To analysis the fruit and vegetable juices for the constituent present in them Viva Voce 7 Rate of Evaporation 1 To study the rate of evaporation of different liquids Viva Voce 8 Effect of Acids and Bases on Tensile Strength of Fibres 1 To compare the tensile strength of natural fibres and synthetic fibres 2 To study the effect of acids and bases on tensile strength of different fibres Viva Voce

General

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