

1. The theory that explains the behavior of gases at the molecular level is called the _____, which is based on assumptions about a theoretical gas often referred to as an _____.
2. Gases deviate most from ideal gas behavior under conditions of very low _____ and very high _____.
 - The molecules of an ideal gas display no _____ or _____ for one another.
 - Under ordinary conditions, an ideal gas consists chiefly of _____ space, which explains why gases are so easily compressed.
 - Ideal gas particles travel in _____ lines until they collide with each other or with the walls of their container.
 - The collisions between the molecules of an ideal gas are completely _____.
 - The average kinetic energy of the molecules of an ideal gas is _____ proportional to the _____ temperature of the gas.
3. A gas exerts pressure on the walls of its container because gas molecules _____ with the walls of the container. So, the pressure exerted by a gas depends on two factors:
 - a)
 - b)
4. To measure gas pressure an instrument called a _____ is used.
5. The earth's atmosphere has weight, which creates _____.
6. The instrument used to measure atmospheric pressure is the _____.

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Lei Huang

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the name 'Lei Huang'.

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