

NAME: _____

CLASS: _____

DATE: _____

1. Which of the following represents an element?

- | | | | |
|-------------------------|--------|-------------------------|----------|
| ☐ A | H_2O | ☐ B | $CaCO_3$ |
| ☐ C | $NaCl$ | ☐ D | H_2 |

2. Which of the following is not a compound?

- | | | | |
|-------------------------|--------|-------------------------|--------|
| ☐ A | CO_2 | ☐ B | $NaCl$ |
| ☐ C | Cl | ☐ D | HCl |

3. Salt has a chemical formula of $NaCl$. Salt is a...

- | | | | |
|-------------------------|----------|-------------------------|-------------------|
| ☐ A | Compound | ☐ B | Element |
| ☐ C | Mixture | ☐ D | None of the Above |

4. What cannot be broken down into other substances?

- | | | | |
|-------------------------|----------|-------------------------|---------|
| ☐ A | Compound | ☐ B | Solids |
| ☐ C | Element | ☐ D | Mixture |

5. What is made up two or more elements that are chemically combined?

- | | | | |
|-------------------------|---------|-------------------------|----------|
| ☐ A | Mixture | ☐ B | Element |
| ☐ C | Solids | ☐ D | Compound |

6. How many elements are represented in the compound?
 Na_2CO_3

- | | | | |
|-------------------------|---|-------------------------|---|
| ☐ A | 2 | ☐ B | 1 |
| ☐ C | 3 | ☐ D | 4 |

7. A combination of substances that are not chemically combined is called

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JS Bruner



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