

$$P_{1}V_{1}T_{2} = P_{2}V_{2}T_{1}$$

$$P_{2} = \frac{P_{1}V_{1}T_{2}}{V_{2}T_{1}}$$

th will be the pressure of the gas if I raise the temperature to 94 °C and rease the volume to 12 liters?

$$P_1 \vee_1 T_2 = P_2 \vee_2 T_1 \qquad P_2 = \frac{(88.89 \pm 1 \text{m})(17 \text{ L})(367 \text{ M})}{(12 \text{ L})(240 \text{ M})}$$

$$P_2 = \frac{P_1 \vee_1 T_2}{\vee_2 T_1} \qquad P_2 = 197.6 \text{ c}^{1}\text{m}$$

18) I have an unknown volume of gas at a pressure of 0.5 atm and a temperature of 325 K. If I raise the pressure to 1.2 atm, decrease the temperature to 320 K, and measure the final volume to be 48 liters, what was the initial volume of the gas?

$$V_1 = \frac{P_2 V_2 T_1}{P_1 T_2}$$

P₁V₁T₂ = P₂V₂T₁ V₁ =
$$\frac{(1.2 \text{ abs})(48 \text{ L})(325 \text{ K})}{(0.5 \text{ abs})(320 \text{ K})}$$
 = 117 L
V₁ = $\frac{P_2 V_2 T_1}{(0.5 \text{ abs})(320 \text{ K})}$ = 277 T

19) The temperature inside my refrigerator is about 4⁶ Celsius. If I place a balloon in my fridge that initially has a temperature of 22°C and a volume of 0.5 liters, what will be the volume of the balloon when it is fully cooled by my refrigerator

$$V_{2} = ?$$
 $295 \frac{1}{15}$
 $V_{1}T_{2} = V_{2}T,$
 $V_{1}T_{2} = V_{2} \cdot \frac{5L \times 277}{275} = .47L$

20) A man heats a balloon in the oven. If the balloon initially has a volume of 0.4 liters and a temperature of 20 °C, what will the volume of the balloon be after he heats it to a temperature of 250 °C?

$$V_1 T_2 = V_2 T_1$$
 $V_2 = (0.4 L)(523 K)$ $V_2 = \frac{V_1 T_2}{T_1}$ $V_2 = 0.71 L$

On hot days, you may have noticed that potato chip bags seem to "inflate", even though they have not been opened. If I have a 250 mL bag at a temperature of 19 OC, and I leave it in my car which has a temperature of 60°C, what will the new volume of the bag be?

$$V_1 T_2 = V_2 T_1$$

$$V_2 = \frac{V_1 T_2}{T_1}$$

$$V_3 = \frac{(250 \text{ mL})(333 \text{ W})}{(292 \text{ W})} = 284.1 \text{ L}$$

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Judd E. Hollander

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