

# Pearson Edexcel IGCSE Chemistry Syllabus (4CH1)

## Paper 2C (Wednesday 15<sup>th</sup> June 2022)

[The topics **highlighted in yellow** have been identified as high priority whereas the others topics will be examined but will have a fewer number of marks attached]

### 1 Principles of Chemistry

#### (d) The Periodic Table

- 1.18 understand how elements are arranged in the Periodic Table: in order of atomic number, in groups and periods
- 1.19 understand how to deduce the electronic configurations of the first 20 elements from their positions in the Periodic Table
- 1.20 understand how to use electrical conductivity and the acid-base character of oxides to classify elements as metals or non-metals
- 1.21 identify an element as a metal or a non-metal according to its position in the Periodic Table
- 1.22 understand how the electronic configuration of a main group element is related to its position in the Periodic Table
- 1.23 understand why elements in the same group of the Periodic Table have similar chemical properties
- 1.24 understand why the noble gases (Group 0) do not readily react

#### (e) Chemical Formulae, Equations and Calculations

- 1.25 write word equations and balanced chemical equations (including state symbols): for reactions studied in this specification and for unfamiliar reactions when suitable information is provided
- 1.26 calculate relative formula masses (including relative molecular masses) ( $M_r$ ) from relative atomic masses ( $A_r$ )
- 1.27 know that the mole (mol) is the unit for the amount of a substance
- 1.28 understand how to carry out calculations involving amount of substance, relative atomic mass ( $A_r$ ) and relative formula mass ( $M_r$ )
- 1.29 calculate reacting masses using experimental data and chemical equations
- 1.30 calculate percentage yield

# Igcse Edexcel Syllabus Chemistry 2014 March

**Kathleen Armour**



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