

Chemistry 100: Final Exam Questions

Question 1: Which compound is a tertiary alcohol?  
 a) Methanol  
 b) Ethanol  
 c) Propan-1-ol  
 d) Propan-2-ol  
 e) Butan-1-ol

Question 2: Which of the following is a saturated hydrocarbon?  
 a) Ethane  
 b) Ethene  
 c) Ethyne  
 d) Propanone  
 e) Propanal

Question 3: Which of the following reactions does not involve a primary alcohol?  
 a) Oxidation of ethanol to ethanal  
 b) Oxidation of propan-1-ol to propanal  
 c) Oxidation of butan-1-ol to butanal  
 d) Oxidation of propan-2-ol to propanone  
 e) Oxidation of ethanol to ethanoic acid

Question 4: A gas of unknown identity has a mass of 0.44 g and a volume of 100 cm<sup>3</sup> at 100 kPa and 273 K. Which of the following is the gas?  
 a) Methane  
 b) Ethane  
 c) Propane  
 d) Butane  
 e) Pentane

Question 5: The compound with a molar mass of 100 g/mol, which is a saturated hydrocarbon, has the formula:  
 a) C<sub>4</sub>H<sub>10</sub>  
 b) C<sub>5</sub>H<sub>12</sub>  
 c) C<sub>6</sub>H<sub>14</sub>  
 d) C<sub>7</sub>H<sub>16</sub>  
 e) C<sub>8</sub>H<sub>18</sub>

Question 6: A sample of pure ethanol (C<sub>2</sub>H<sub>5</sub>OH) has a mass of 10.0 g. How many moles of ethanol are present?  
 a) 0.22 mol  
 b) 0.44 mol  
 c) 0.66 mol  
 d) 0.88 mol  
 e) 1.10 mol

Question 7: Which of the following reactions does not involve a redox reaction?  
 a) Oxidation of ethanol to ethanal  
 b) Oxidation of propan-1-ol to propanal  
 c) Oxidation of butan-1-ol to butanal  
 d) Oxidation of propan-2-ol to propanone  
 e) Oxidation of ethanol to ethanoic acid

Question 8: The reaction of ethene with steam to form ethanol is an exothermic reaction. Which of the following is a correct statement about this reaction?  
 a) The reaction is reversible.  
 b) The reaction is irreversible.  
 c) The reaction is a redox reaction.  
 d) The reaction is a non-redox reaction.  
 e) The reaction is a combustion reaction.

Question 9: The reaction of ethene with steam to form ethanol is an exothermic reaction. Which of the following is a correct statement about this reaction?  
 a) The reaction is reversible.  
 b) The reaction is irreversible.  
 c) The reaction is a redox reaction.  
 d) The reaction is a non-redox reaction.  
 e) The reaction is a combustion reaction.

Question 10: A sample of pure ethanol (C<sub>2</sub>H<sub>5</sub>OH) has a mass of 10.0 g. How many moles of ethanol are present?  
 a) 0.22 mol  
 b) 0.44 mol  
 c) 0.66 mol  
 d) 0.88 mol  
 e) 1.10 mol

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## **Table of Contents Grossmont College Assessment Practice Test**

1. Understanding the eBook Grossmont College Assessment Practice Test
  - The Rise of Digital Reading Grossmont College Assessment Practice Test
  - Advantages of eBooks Over Traditional Books
2. Identifying Grossmont College Assessment Practice Test
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Grossmont College Assessment Practice Test
  - User-Friendly Interface
4. Exploring eBook Recommendations from Grossmont College Assessment Practice Test
  - Personalized Recommendations
  - Grossmont College Assessment Practice Test User Reviews and Ratings
  - Grossmont College Assessment Practice Test and Bestseller Lists
5. Accessing Grossmont College Assessment Practice Test Free and Paid eBooks
  - Grossmont College Assessment Practice Test Public Domain eBooks
  - Grossmont College Assessment Practice Test eBook Subscription Services
  - Grossmont College Assessment Practice Test Budget-Friendly Options

6. Navigating Grossmont College Assessment Practice Test eBook Formats
  - ePub, PDF, MOBI, and More
  - Grossmont College Assessment Practice Test Compatibility with Devices
  - Grossmont College Assessment Practice Test Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Grossmont College Assessment Practice Test
  - Highlighting and Note-Taking Grossmont College Assessment Practice Test
  - Interactive Elements Grossmont College Assessment Practice Test
8. Staying Engaged with Grossmont College Assessment Practice Test
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Grossmont College Assessment Practice Test
9. Balancing eBooks and Physical Books Grossmont College Assessment Practice Test
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Grossmont College Assessment Practice Test
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Grossmont College Assessment Practice Test
  - Setting Reading Goals Grossmont College Assessment Practice Test
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Grossmont College Assessment Practice Test
  - Fact-Checking eBook Content of Grossmont College Assessment Practice Test
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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