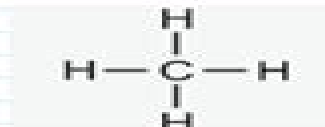


Organic chemistry Page 29

- A homologous series are a group of organic compounds with similar chemical properties because of the same functional group.
- The functional group is an atom or group of atoms that give a molecule its characteristic properties.

Structural formulae of methane



Some facts to know

Suffix	compound	Functional group
ane	Alkane	none
ene	Alkene	Carbon carbon double bonds
anol	alcohol	OH
anoic acid	Carboxylic acid	COOH
Yl/ethanoate	Ester	COO

- Straight-chain hydrocarbons have carbon atoms bonded with two or fewer carbon atoms. A hydrocarbon becomes branched if any carbon atoms form more than two bonds with other carbon atoms.

Homologous series have the following properties,

- Having the same functional group.
- Same general formula.
- Differing from one member to the next by a $-\text{CH}_2-$ unit.
- Displaying a trend in chemical properties.
- Having similar chemical properties.

Homologous series	Formula
Alkanes	$\text{C}_n\text{H}_{2n+2}$
Alkenes	C_nH_{2n}
Alcohol	$\text{C}_n\text{H}_{2n+1}\text{OH}$
Carboxylic acid	$\text{C}_n\text{H}_{2n+1}\text{COOH}$

- The Saturated compound has molecules in which all carbon-carbon bonds are single.
- An unsaturated compound has molecules in which one or more carbon-carbon bonds are not single bonds.
- Structural isomers are compounds with the same molecular formulae but different structural formulas.

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