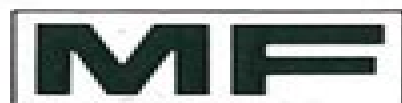


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extreme exploits as they strive to stay afloat They run legal when they can and cut corners when necessary gambling on the new Scandinavian trucks that are changing the industry They are cheated out of their earnings but still come up smiling

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web october 3 2017 abstract the techniques used in this lab are quantitative transfer and vacuum filtration with the reaction

of 8.001 grams of copper ii sulfate  $\text{CuSO}_4$  and 2.0153 grams of iron powder  $\text{Fe}$  the goal of this experiment was to determine the product of copper ii sulfate with iron

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web procedure 1 find the mass of a clean dry 100 ml beaker and record 2 measure out 8.00 grams of copper ii sulfate onto a piece of filter paper taking the filter paper's mass into account add these to the beaker 3 measure 50.0 ml of water in the cylinder and add it to the beaker 4

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web description great lab for use during stoichiometry unit in chemistry reinforces ideas of mole ratios and stoichiometry calculations by measuring the mass of iron that reacts and the mass of copper metal produced students can calculate the ratio of moles of reactant to moles of product this experimental mole ratio from lab can be compared

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web under the finalise of the experiment it was found that iron was the limiting reactant and iron ii sulfate was a product of the reaction between 7 0002g away copper ii sulfate and 2 0101g iron metal which produced 2 3037g of copper product with a percent yield of 100 72 introduction

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imagine that you have an object suspended by transparent threads inside a glass box as in figure 3 figure 3 the block suspended in a glass box then draw the object on each of three faces as seen from that see more

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figure 7 an isometric view with dimensions we have dimensioned the object in the isometric drawing in figure 7 as a general guideline to dimensioning try to think that you would make an object and dimension it in the see more

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