

**Guided Reading Activity 17-1****The Scientific Revolution**

DIRECTIONS: Answer the following questions as you read Section 1.

1. What did the writings of Ptolemy and Archimedes make obvious?

2. What new invention helped to spread new scientific ideas quickly and easily?

3. Where is Earth placed in the universe according to the Ptolemaic system?

4. Contrary to Ptolemy, what did Copernicus argue concerning the construction of the universe?

5. What discoveries did Galileo make using a telescope?

6. Why did the Church order Galileo to abandon the Copernican idea of the nature of the universe?

7. What did Isaac Newton define in his first book, *The Principia*?

8. What did William Harvey's observations and experiments show?

9. What field of science in Germany provided opportunities for women?

10. What did René Descartes emphasize and assert?

11. Who developed the scientific method?

Guided Reading Activity 17 1 The Scientific Revolution

Answer Key

WJ Hussar



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1. This book is structured into several chapters, namely:
 - Chapter 1: Introduction to Guided Reading Activity 17 1 The Scientific Revolution Answer Key
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 - Chapter 3: Guided Reading Activity 17 1 The Scientific Revolution Answer Key in Everyday Life
 - Chapter 4: Guided Reading Activity 17 1 The Scientific Revolution Answer Key in Specific Contexts
 - Chapter 5: Conclusion
 2. In chapter 1, the author will provide an overview of Guided Reading Activity 17 1 The Scientific Revolution Answer Key. This chapter will explore what Guided Reading Activity 17 1 The Scientific Revolution Answer Key is, why Guided Reading Activity 17 1 The Scientific Revolution Answer Key is vital, and how to effectively learn about Guided Reading Activity 17 1 The Scientific Revolution Answer Key.
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- The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Guided Reading Activity 17 1 The Scientific Revolution Answer Key.

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web so basing on my understanding of this so far an atom is an element that has the same number of protons and neutrons

an ion is an element that has different numbers of protons and electrons an ion is positive when it has more protons than electrons and negative when it has more electrons than ions while isotopes are the same element with

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web figure pageindex 2 the formation of a chlorine ion on the left the chlorine atom has 17 electrons on the right the chloride ion has 18 electrons and has a 1 charge with two oppositely charged ions there is an electrostatic attraction between them because opposite charges attract

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