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# House Of Night S In Order

**Victor M. Corman**



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web calculate the mass percent mole fraction molality and molarity of hexane  $C_6H_{14}$   $d = 0.660 \text{ g/cm}^3$  when it is mixed with heptane  $C_7H_{12}$   $d = 0.684 \text{ g/cm}^3$  in a 35.0 : 55.0 ml ratio assume that the volumes are additive

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web since the mole fraction of the  $\text{O}_2$  is  $0.050$  it contributes  $0.050$  or if you will  $5.0\%$  of the total pressure thus total pressure is  $0.2105042 \text{ atm}$  problem 9 a sample of  $1.43 \text{ g}$  of helium and an unweighed quantity of  $\text{O}_2$  are mixed in a flask at room temperature