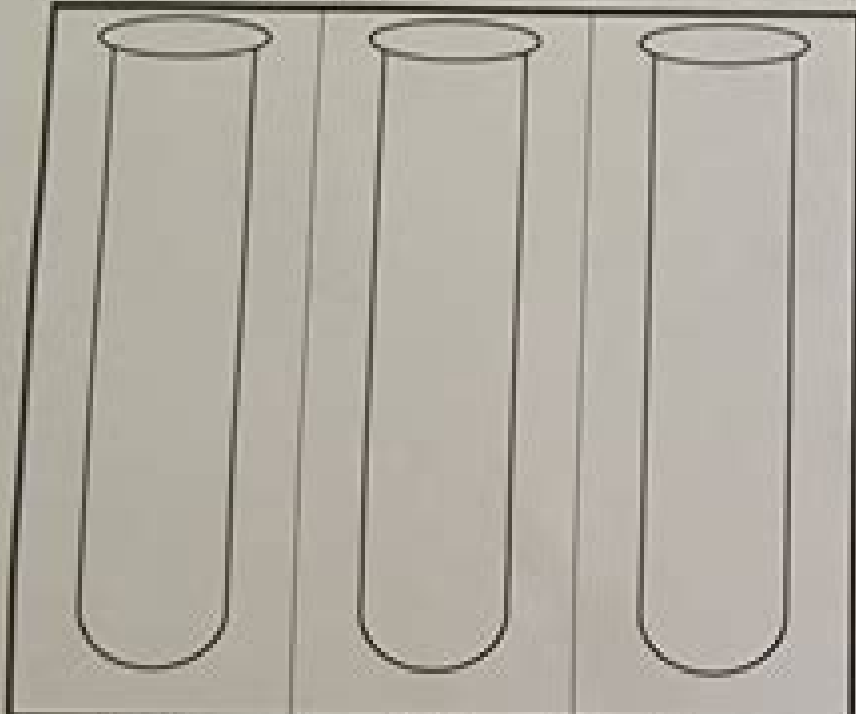


# Chemistry 162 Nine Solutions Problem – Post Lab (Upload as an image to Canvas)

For each set of reactants shown: 1) Create small particle drawings of the separate reactant solutions and the solution after mixing. Maintain stoichiometry and include at least 2 H<sub>2</sub>O molecules for (aq) mixtures 2) Write the balanced molecular reaction. 3) Write the balanced complete ionic equation (no cross-outs). 4) Write the balanced net ionic equation. Include phases for all species. See the lab manual for an example.



AgNO<sub>3</sub>

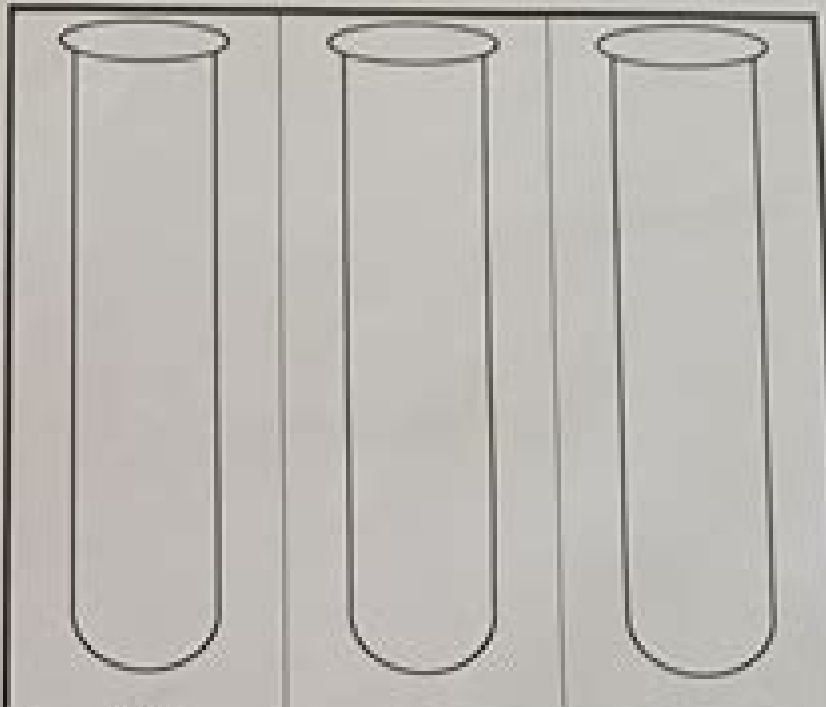
Na<sub>2</sub>SO<sub>4</sub>

After mixing

Balanced chemical equation:

Complete ionic equation

Net ionic equation



BaCl<sub>2</sub>

NaBr

After mixing

Balanced chemical equation:

Complete ionic equation

Net ionic equation

# Hayden Mcneil Lab Manual Answers Chem 112

**Rosina Ehmann**



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