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# KENWOOD

Kenwood Corporation

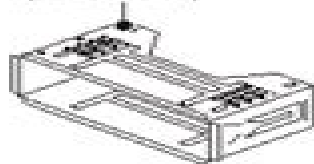
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B53-0513-00 (N) 552



## SPARE TDF PANEL

MAIN UNIT NAME	TDF PARTS No.	TDF NAME
KDC-MP4036U	Y33-2720-64	TDF-MP4036U
KDC-MP435U	Y33-2720-61	TDF-MP740
KDC-W4537UA	Y33-2720-63	TDF-W4537U
KDC-W4537UAY	Y33-2720-63	TDF-W4537U
KDC-W4537UG	Y33-2720-63	TDF-W4537U
KDC-W4537UGY	Y33-2720-63	TDF-W4537U
KDC-W4737U	Y33-2720-62	TDF-W4737U
KDC-W4737UY	Y33-2720-62	TDF-W4737U
KDC-X491	Y33-2720-60	TDF-MP740X

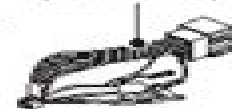
Mounting hardware Assy  
(J21-9715-03)



\* DC cord  
(E30-6425-05)



\* DC cord  
(E30-6571-05)



\* Remote controller Assy (RC-xxx)  
(A70-xxxx-xx)



\* Antenna adaptor  
(T90-0523-05)



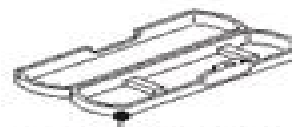
\* Screw set  
(N99-1757-05)



Lever  
(D10-4529-04) x2



\* Plastic cabinet Assy  
(A02-2735-03)



\* Carrying case  
(W01-xxxx-xx)



Escutcheon  
(B07-3125-01)



\* Depends on the model. Refer to the parts list.

This product uses Lead Free solder.

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# Kenwood Kdc X491 Manual

**D Siedentop**



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type of chemical reaction for this reaction we have a neutralization reaction balancing strategies here we have a neutralization reaction

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web the balanced chemical equation for this reaction is  $\text{mg} + 2\text{hno}_3(\text{aq}) \rightarrow \text{mg}^{2+} + 2\text{no}_3^- + \text{h}_2(\text{g})$  in this equation the s symbolizes solid magnesium aq represents aqueous nitric acid aq signifies magnesium nitrate in a dissolved state and g represents the gaseous hydrogen produced

[\$\text{hno}\_3 + \text{mg} + \text{c}\_2\text{h}\_3\text{o}\_2 \rightarrow 2\text{hc}\_2\text{h}\_3\text{o}\_2 + \text{mg}^{2+} + \text{no}\_3^-\$  chemical equation](#) - Aug 11 2023

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web mar 4 2018  $\text{mg} + 2\text{hno}_3 \rightarrow \text{mg}^{2+} + \text{no}_3^- + \text{h}_2$  on the left side there are mg 1 magnesium in elemental form  $\text{h}_2$  there are 2 molecules of nitric acid each acid has one hydrogen so two molecules of nitric acid is 2 hydrogens n 2 there are 2 molecules of nitric acid each acid has one nitrogen so two molecules of nitric acid is 2 nitrogens o 6

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web magnesium carbonate nitric acid magnesium nitrate water carbon dioxide one mole of magnesium carbonate  $\text{mgco}_3$  and two moles of nitric acid  $\text{hno}_3$  react to form one mole of magnesium nitrate  $\text{mg no}_3 2$  one mole of water  $\text{h}_2\text{o}$  and one mole of carbon dioxide  $\text{co}_2$  show chemical structure image reactants

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web magnesium is already balanced on the reactant and product side balancing nitrogen by multiplying  $\text{hno}_3$  by 2 we get  $\text{mg s}$  magnesium 2  $\text{hno}_3$  l nitric acid  $\text{mg no}_3 2$  s magnesium nitrate  $\text{h}_2\text{g}$  hydrogen gas oxygen and hydrogen get automatically balanced by balancing nitrogen

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each compound to balance  $\text{mg hno}_3$   $\text{mg no}_3$   $2 \text{ no}_2$   $\text{h}_2\text{o}$  you ll need to be

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$\text{mg} 2 2 \text{ oh} 1 2 \text{ h} 1 2 \text{ no}_3 1 2 \text{ h}_2 \text{ o}$   $\text{mg} 2 2 \text{ no}_3 1$  net ionic equation  $2 \text{ oh} 1 2 \text{ h} 1 2 \text{ h}_2 \text{ o}$  sulfuric acid and lithium hydroxide

balanced total equation  $\text{h}_2 \text{ so}_4 \text{ aq} 2 \text{ lihoh aq} 2 \text{ h}_2$

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