

Heat and Other Forms of Energy

5-101 The rate of heat transfer per unit surface area is called heat flux, \dot{q} . It is related to the rate of heat transfer by $\dot{Q} = \int \dot{q} dA$.

5-102 Energy can be transferred by heat, work, and mass. An energy transfer is heat transfer when the driving force is temperature difference.

5-103 Thermal energy is the sensible and latent forms of internal energy, and it is related to us heat in daily life.

5-104 For the constant pressure case, this is because the heat transfer to an ideal gas is $m c_p \Delta T$ at constant pressure and $m c_v \Delta T$ at constant volume, and c_p is always greater than c_v .

5-122 A cylindrical resistor on a circuit board dissipates 0.6 W of power. The amount of heat dissipated in 24 h, the heat flux, and the fraction of heat dissipated from the top and bottom surfaces are to be determined.

Assumptions Heat is transferred uniformly from all surfaces.

Analysis (a) The amount of heat the resistor dissipates during a 24-hour period is

$$\dot{Q} = \dot{Q}_{\text{res}} = (0.6 \text{ W})(24 \text{ h}) = 14.4 \text{ Wh} = 51.8 \text{ kJ} \quad \text{since } 1 \text{ Wh} = 3600 \text{ Wh} = 3.6 \text{ kJ}$$

(b) The heat flux on the surface of the resistor is

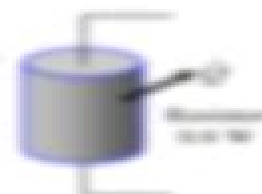
$$\dot{q}_s = \frac{\dot{Q}_s}{A_s} = \frac{0.6 \text{ W}}{A_s} = \frac{0.6 \text{ W}}{\frac{\pi D^2}{4}} = \frac{0.6 \text{ W}}{\frac{\pi (0.002 \text{ m})^2}{4}} = 6,370 \text{ W/m}^2 = 6.370 \times 10^3 \text{ W/m}^2$$

$$\dot{q}_t = \frac{\dot{Q}_t}{A_t} = \frac{0.60 \text{ W}}{2 \times 1.768 \text{ cm}^2} = 0.339 \text{ W/cm}^2$$

(c) Assuming the heat transfer coefficient to be uniform, heat transfer is proportional to the surface area. Then the fraction of heat dissipated from the top and bottom surfaces of the resistor becomes

$$\frac{\dot{Q}_{\text{top, bot}}}{\dot{Q}_{\text{res}}} = \frac{A_{\text{top, bot}}}{A_{\text{res}}} = \frac{0.000177 \text{ m}^2}{1.768 \text{ cm}^2} = 0.0099 \quad \text{or } 0.99\%$$

Discussion Heat transfer from the top and bottom surfaces is small relative to that transferred from the side surface.



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Introduction to Heat Transfer. Solution Manual D.P. Dewitt, 1996 **Heat Transfer S/M Sup** Anthony F. Mills, Mills, 1998-10 **Solutions Manual to Accompany Heat Transfer, Seventh Edition** Jack Philip Holman, 1990 Engineering Heat Transfer William S. Janna, 2018-10-03 Most heat transfer texts include the same material conduction convection and radiation How the material is presented how well the author writes the explanatory and descriptive material and the number and quality of practice problems is what makes the difference Even more important however is how students receive the text Engineering Heat Transfer Third Edition provides a solid foundation in the principles of heat transfer while strongly emphasizing practical applications and keeping mathematics to a minimum New in the Third Edition Coverage of the emerging areas of microscale nanoscale and biomedical heat transfer Simplification of derivations of Navier Stokes in fluid mechanics Moved boundary flow layer problems to the flow past immersed bodies chapter Revised and additional problems revised and new examples PDF files of the Solutions Manual available on a chapter by chapter basis The text covers practical

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