

## HALF-LIFE PROBLEMS

Name \_\_\_\_\_ Block \_\_\_\_\_

1. An isotope of cesium (cesium-137) has a half-life of 30 years. If 1.0 g of cesium-137 disintegrates over a period of 90 years, how many g of cesium-137 would remain?

$$\frac{90}{50} = 3 \frac{1}{2}$$

$\begin{pmatrix} 1.5 \\ 0.5 \\ 2.5 \\ 12.5 \text{ g} \end{pmatrix}$

2. Actinium-226 has a half-life of 29 hours. If 100 mg of actinium-226 disintegrates over a period of 58 hours, how many mg of actinium-226 will remain?

58-2-41

100  
50  
25

3. Sodium-25 was to be used in an experiment, but it took 3.0 minutes to get the sodium from the reactor to the laboratory. If 5.0 mg of sodium-25 was removed from the reactor, how many mg of sodium-25 were placed in the reaction vessel 3.0 minutes later if the half-life of sodium-25 is 60 seconds?

$$\frac{3.0}{1.0} = 34.1$$

$\begin{cases} 5 \text{ mg} \\ 2.5 \\ 1.25 \\ 0.625 \text{ mg} \end{cases}$

4. The half-life of isotope X is 2.0 years. How many years would it take for a 4.0 mg sample of X to decay and have only 0.50 mg of it remain?

$$3 \times 2.0 \text{ yr} = 6 \text{ yr}$$

4.0  
2.0  
1.0  
5.0 } 3 h.L

5. Selenium-83 has a half-life of 25.0 minutes. How many minutes would it take for a 10.0 mg sample to decay and have only 1.25 mg of it remain?

$$3 \times 25 = 75 \text{ min}$$

1.25 3 h.L

6. The half-life of Po-218 is three minutes. How much of a 2.0 gram sample remains after 15 minutes? Suppose you wanted to buy some of this isotope, and it required half an hour for it reach you. How much should you order if you need to use 0.10 gram of this material?

15. 2nd

2.3  
1.5  
2.2  
1.25  
1.0425

300, 10 h.

$10 \text{ h.l}$

$\cdot 1 \times 2 \times 2 \times 2 \times 2 \quad 2 \times 2 \times 2 = 2 + 2 \times 2$

$102.4g$

# Half Life Calculations Physical Science If8767

**RS Peters**



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## **Table of Contents Half Life Calculations Physical Science If8767**

1. Understanding the eBook Half Life Calculations Physical Science If8767
  - The Rise of Digital Reading Half Life Calculations Physical Science If8767
  - Advantages of eBooks Over Traditional Books
2. Identifying Half Life Calculations Physical Science If8767
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Half Life Calculations Physical Science If8767
  - User-Friendly Interface
4. Exploring eBook Recommendations from Half Life Calculations Physical Science If8767
  - Personalized Recommendations
  - Half Life Calculations Physical Science If8767 User Reviews and Ratings
  - Half Life Calculations Physical Science If8767 and Bestseller Lists

5. Accessing Half Life Calculations Physical Science If8767 Free and Paid eBooks
  - Half Life Calculations Physical Science If8767 Public Domain eBooks
  - Half Life Calculations Physical Science If8767 eBook Subscription Services
  - Half Life Calculations Physical Science If8767 Budget-Friendly Options
6. Navigating Half Life Calculations Physical Science If8767 eBook Formats
  - ePub, PDF, MOBI, and More
  - Half Life Calculations Physical Science If8767 Compatibility with Devices
  - Half Life Calculations Physical Science If8767 Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Half Life Calculations Physical Science If8767
  - Highlighting and Note-Taking Half Life Calculations Physical Science If8767
  - Interactive Elements Half Life Calculations Physical Science If8767
8. Staying Engaged with Half Life Calculations Physical Science If8767
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Half Life Calculations Physical Science If8767
9. Balancing eBooks and Physical Books Half Life Calculations Physical Science If8767
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Half Life Calculations Physical Science If8767
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Half Life Calculations Physical Science If8767
  - Setting Reading Goals Half Life Calculations Physical Science If8767
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Half Life Calculations Physical Science If8767
  - Fact-Checking eBook Content of Half Life Calculations Physical Science If8767
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

#### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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