# Chapter 9 Study Guide Holt Chemistry: Stoichiometry

# Questions

- What is stoichiometry? Finding mass of product formed for reactant consumed) when given a mass of another species in the reaction, by using the mole to mole ratio from the balanced equation
- 2. A balanced equation verifies the law of conservation of mass. Total mass of reactuats must equal total mass of products.
- The Clive the 3 steps of converting the mass (grams) of one substance to the mass (grams) of another substance: It. Convert grams to moles using molar mass; 2. Mole to Mole ratio from balanced equation; 3. Convert moles back to grams using molar mass

# **Holt Chemistry Study Guide Stoichiometry Answer Key**

**S Marginson** 

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